Cycloaddition of Fluorenone N‑Aryl Nitrones with Methylenecyclopropanes and Sequential 1,3-Rearrangement: An Entry to Synthesis of Spirofluorenylpiperidin-4-ones

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S Supporting Information

[AB](#page-8-0)STRACT: [A facile synth](#page-8-0)esis of various spirofluorenylpiperidin-4-ones has been achieved in good yields from fluorenone N-aryl nitrones and methylenecyclopropanes. This method involved an initial cycloaddition to form a 5-spirocyclopropane-isoxazoline, which underwent a highly selective 1,3 rearrangement to give the desired product. The stereochemistry of the spirofluorenylpiperidin-4-one could be

controlled by the cycloaddition and sequential rearrangement strategy. Furthermore, the spirofluorenylpiperidin-4-ones could be not only prepared in one-pot procedure but also converted to useful scaffolds by reduction or oxidation conditions.

ENTRODUCTION

The development of new and efficient methods for synthesis of piperidinone derivatives represents an important research field because these piperidinone scaffolds showed the unique biological activities and served as valuable intermediates in organic synthesis.^{1,2} 1,3-Dipolar cycloaddition of nitrones to the double bond of methylenecyclopropanes is one of the most extensively studi[ed](#page-9-0) cycloaddition reactions due to the rich chemistry of 5-spirocyclopropane-isoxazoline cycloadducts formed in the reactions. $3,4$ 5-Spirocyclopropane-isoxazolines could undergo Brandi−Guarna rearrangement to afford piperidin-4-ones or relate[d c](#page-9-0)ompounds because of the highly strained spirocyclopropane ring.⁵ In 1989, Brandi and coworkers found that the reaction of (Z) -N-aryl aldonitrones and methylenecyclopropane could f[or](#page-9-0)m 4-spirocyclopropane-isoxazoline and 5-spirocyclopropane-isoxazoline which could be converted to a mixture of enamine, piperidin-4-one, and benzoazocinone (Scheme $1A$).⁶ A further study on the regioselective formation of 5-spirocyclopropane-isoxazoline and its 1,3- or 3,3-rearrangemen[t](#page-9-0) was reported by Molchanov and co-workers (Scheme 1B).⁷ The effect of substituted groups on both nitrones and methylenecyclopropanes to the N−O bond rearrangement was stu[di](#page-9-0)ed. Although the rearrangement of 5-spirocyclopropane-isoxazoline has been studied, the yields and selectivities observed in the 1,3- and 3,3-rearrangement of 5-spirocyclopropane-isoxazoline are generally poor for N-aryl nitrones. During the studies on the rearrangement of N−O bond in our group, δ we envisaged that methylenecyclopropanes bearing an alkoxycarbonyl group would regioselectively react with N-aryl nitron[es](#page-9-0) bearing a rigid and sterically encumbered 9-fluorenyl group to give 5-spirocyclopropane-isoxazolines, which might undergo a controllable rearrangement process.

Scheme 1. Rearrangement of N−O Cyclopropane Intermediates

A) Brandi' work, mixture of [3,3]- and [1,3]-rearrangement

B) Molchanov's work, substituted groups effect on [3,3]- and [1,3]-rearrangement

C) This work, high selective [1,3]-rearrangement

Fluorenone nitrones are distinct from traditional nitrones because of the rigidity and steric hindrance of the fluorene.⁹ This property can be exploited by N–O bond cleavage¹⁰ to

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afford spirofluorenylpiperidinones that exhibit privileged scaffolds present in bioactive pharmaceuticals and organic materials.¹¹ Herein, we report a facile synthesis of spirofluorenylpiperidin-4-ones in good yields from fluorenone nitrones and met[hyl](#page-9-0)enecyclopropanes (MCP) (Scheme 1C).

■ RESULTS AND DISCUSSION

We initially evaluated the formation [of](#page-0-0) [5-spiro](#page-0-0)cyclopropaneisoxazoline via the cycloaddition reaction of N-aryl fluorenone nitrone and methylenecyclopropane. N-Phenyl fluorenone nitrone 1a and methylenecyclopropane 2a were chosen as model substrates for this reaction. As shown in Table 1, the

Table 1. Optimization of the Cycloaddition Reaction Conditions α

$Ph\sqrt{N}$ 1a	2a	Ph Ph. conditions	Ph 'N ∖∖ O 3a
	solvent	$T({}^{\circ}C)$	3a yield $(\%)^b$
$\mathbf{1}$	toluene	40	98
$\overline{2}$	DCE	40	98
3	MeCN	40	96
$\overline{4}$	THF	40	97
5	DMSO	40	79
6	MeOH	40	94
7	DCE	25	99 ^c
8	DCE	60	84
9	DCE	80	$<$ 5
σ -	\prime . \sim \sim	\bullet λ ϵ .	\sim

^aReaction conditions: 1a (0.2 mmol) , 2a $(0.4 \text{ mmol}, 2.0 \text{ equiv})$, solvent (2 mL) , 18−24 h. $\frac{b_{12}}{2}$ inners, $\frac{c_{13}}{2}$ calcion time: 36 h.

desired cycloadduct 3a was isolated in excellent yield from a variety of different solvents, including toluene, 1,2-dichloroethane (DCE), MeCN, THF, DMSO, and MeOH following 18−24 h at 40 °C (Table 1, entries 1−6). Reducing the reaction temperature to room temperature led to an increase in the yield of 3a to 99%, albeit after extending the reaction time to 36 h (Table 1, entry 7). In contrast, increasing the reaction temperature to 60 °C led to a decrease in the yield of 3a (Table 1, entry 8). Furthermore, no product 3a was observed when the reaction was conducted at 80 °C, with the starting materials 1a and 2a being recovered unchanged (Table 1, entry 9).

The cycloaddition reactions of various N-aryl fluorenone nitrones 1 with methylenecyclopropane 2a were conducted under the optimal conditions. As shown in Table 2, these reactions proceeded smoothly to afford the desired products in good to excellent yields. This reaction was found to be compatible with several useful functional groups on the nitrone substrates, including methoxy, bromide, ester, and nitro groups at the para, ortho, or meta-position. However, nitrones 1b and 1k bearing a para-methoxy and ortho-bromide, respectively, afforded low yields of the corresponding cycloadducts 3b and 3k (Table 2, entries 2 and 11). Notably, nitrone 1m with a 1:1 ratio of E/Z -isomer gave product 3m with a dr value of 1:1 when it reacted with 2a under the optimal conditions (Table 2, entry 13).

In addition to the nitrone moiety, we also investigated a variety of different methylenecyclopropane substrates 2a−j to determine the substituent effect on the cycloaddition reaction. All of these compounds provided the desired cycloadduct 3 in Table 2. Scope of the Nitrones 1^a

a Reaction conditions: 1 (0.2 mmol), 2a (0.4 mmol, 2.0 equiv), solvent (2 mL), $18-36$ h. $\frac{b}{b}$ Isolated yield. $\frac{c}{c}$ dr value.

good to excellent yields (Table 3). Notably, when (−)-menthol methylenecyclopropane 2h was evaluated as a substrate to explore the diastereosel[ectivity o](#page-2-0)f the cycloaddition reaction, the corresponding product 3ah was isolated in 58% yield with a dr value of 3.5:1, which was determined by ${}^{1}H$ NMR (Table 3, entry 8). This result revealed that the stereochemistry of the cycloadduct could be controlled by installing a chiral au[xiliary in](#page-2-0) the methylenecyclopropane substrate. The ketone-substituted methylenecyclopropane 2i gave the corresponding cycloadduct 3ai in 73% yield (Table 3, entry 9). However, the reaction of methylenecyclopropane 2j bearing a primary alcohol failed to afford the desire[d cyclo](#page-2-0)adduct 3aj, and nitrone 1a and methylenecyclopropane 2j were recovered unchanged (Table 3, entry 10).

Having fully evaluated the scope of the cycload[dition](#page-2-0) [re](#page-2-0)action, we proceeded to investigate the rearrangement of 5 spirocyclopropane-isoxazoline 3. When 3a was heated at 100 °C, the nitrone 1a was recovered with 35% yield; N-(9Hfluoren-9-ylidene)aniline was isolated in 20% yield, and the enolated 1,3-rearrangement product 4a′ was obtained in 10% yield (Scheme 2-1). This experiment revealed that the cycloaddition reaction might be reversible, and the desired selectiv[e rearrangem](#page-2-0)ent could be achieved through improving the rigidity of cycloaddition intermediate. To inhibit the reversible reaction and promote the efficiency of the rearrangement process, the 5-spirocyclopropane-isoxazoline 3a was reduced by $LiAlH₄$ to provide alcohol 3aj. When 3aj was heated at 40 °C for 36 h, no desired product 4a was observed, and only starting material 3aj was recovered. To our delight, only 1,3-rearrangement product 4a was obtained in 64% yield while 3aj was heated at 80 °C for 18 h, and the solvent played an important role on the yield of product 4a (Scheme 2-2). The optimal conditions for the rearrangement of 3aj to 4a were in MeCN at 120 °C.

With the optimal conditions, we found tha[t](#page-2-0) [the](#page-2-0) [redu](#page-2-0)ction of 3a with LiAlH₄, followed by the rearrangement of the resulting alcohol in MeCN at 120 °C afforded product 4a in 70% isolated yield in two steps. As shown in Table 4, the

Table 3. Scope of the Methylenecyclopropane Substrate 2^a

a
Reaction conditions: 1a (0.2 mmol), 2 (0.4 mmol, 2.0 equiv), DCE (2 mL), 18–36 h. $\frac{b}{b}$ Isolated yield. ^cdr value.

rearrangement products 4 were obtained in yields of 46−87% from the corresponding cycloadduct 3 by sequential reduction and rearrangement reactions (method A) or sequential reduction, protection (with an alkyl bromide) and rearrangement reactions (method B) in a single-flask. The reaction was compatible with electron-withdrawing and electron-donating groups on the aryl rings. Furthermore, 3h containing two ester groups reacted smoothly to give product 4h in 73% yield bearing two hydroxyl groups (Table 4, entry 7), which indicated that the two ester groups of 3h were reduced by $LiAlH₄$.

The stereochemistry of the cycloaddition and rearrangement steps was investigated using the reaction shown in Scheme 3. Reduction of cycloadduct $3ah$ with $LiAlH₄$ afforded the corresponding alcohol 3aj in 86% yield and 35% [ee value](#page-3-0). The desired rearrangement product 4a was obtained in 89% yield and 37% ee value when chiral 3aj was subjected to the optimal conditions. This result suggested that the absolute stereochemistry of the spiropiperidin-4-one could be controlled through this cycloaddition and rearrangement process.

The types of nitrones were also examined to the cycloaddition and rearrangement process (Scheme 4). When fluorenone N-methyl nitrone 1n reacted with methylenecyclopropane 2a in DCE at room temperature f[ollowed by](#page-3-0) reduction

1) Thermal reaction of cycloadduct 3a

method B: 1) LiAlH₄; 2) R⁴Br, Et₃N, 3) MeCN, 120 °C

^aReaction conditions: method A: (1) 3 (0.2 mmol), LiAlH₄ (2.0) equiv), THF, 0 °C, 30 min; (2) MeCN (2 mL), 120 °C, 18−24 h. method B: (1) 3 (0.2 mmol), LiAlH₄ (2.0 equiv), THF, 0 °C, 10-30 min; (2) R⁴Br (2.0 equiv), 2–10 h; (3) MeCN (2 mL), 120 °C, 18– 24 h. b^b Isolated yield. ^cThe methoxycarbonyl group in 3h was also $r = \frac{1}{2}$ and $r = \frac{d}{dr}$ value.

of LiAlH4, the alcohol 3na was obtained in 95% yield. However, the rearrangement of 3na did not occur in the optimal conditions, and only the alcohol 3na was recovered (Scheme 4- 1). To our surprise, both (Z) -N-phenyl benzaldehyde nitrone

Scheme 3. Stereochemistry in the Rearrangement Step

Scheme 4. Effect of Type of Nitrones and Methylenecyclopropanes on the Cycloaddition and Rearrangement

10 and (Z) -N-methyl benzaldehyde nitrone 1p with 2a could not provide the cycloadducts at room temperature, and the reaction turned messy (Scheme 4-2 and 4-3). When nitrone 1a reacted with methylenecyclopropane 2k at either room temperature or high temperature, no desired cycloadduct or rearrangement product was obtained with starting materials being recovered (Scheme 4-4). The results revealed that the types of nitrones and methylenecyclopropanes are important for these transformations.

To explore the utility of this cascade process, we further studied the efficiency of one-pot process and the usefulness of spiropiperidin-4-one. Pleasingly, spiropiperidin-4-one 4a was prepared in 83% yield in a total of three steps via a one-pot reaction from nitrone 1a and methylenecyclopropane 2a (Scheme 5-1). Reduction of $4a$ with NaBH₄ afforded the corresponding 1,3-diol 5 in 82% yield with a dr value of 20:1. Treatment of 4a with Dess−Martin periodinane (DMP) gave compound 6 in 88% yield.¹² The ability to install useful

Scheme 5. Transformation [Stu](#page-9-0)dies

functional groups on spiropiperidin-4-one will undoubtedly lead to these compounds in more applications.

■ **CONCLUSIONS**

We developed an efficient synthesis of fluorenyl-substituted spiropiperidin-4-ones from N-aryl fluorenone nitrones and methylenecyclopropanes. This reaction involved initial cycloaddition, reduction and sequential thermal rearrangement steps to give spirofluorenylpiperidin-4-ones, and exhibited functional group compatibility toward halide, ester, alcohol, and nitro groups. The 1,3-rearrangement of the N−O bond could be controlled by the rigidity of the fluorenyl in the nitrone. Notably, this reaction allowed for the one-pot synthesis of spirofluorenylpiperidin-4-one, providing a powerful platform for the construction of numerous other diverse scaffolds.

EXPERIMENTAL SECTION

General Methods. All reactions were performed under an atmosphere of air. Commercially available reagents were used without further purification. The NMR spectra were recorded in $CDCI₃$ or DMSO- d_6 on 400 or 500 MHz instrument with TMS as the internal standard. NMR data are represented as follows: chemical shift (ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constants in hertz (Hz), and integration. IR spectra were recorded on FT-IR spectrometer, and only major peaks are reported in cm⁻¹. HRMS were measured in ESI mode and the mass analyzer of the HRMS was TOF. Flash column chromatography was performed on silica gel (300−400 mesh).

General Procedure A for Synthesis of Cycloadduct 3 (Tables 2 and 3). In a Teflon-sealed reaction flask charged with nitrone 1 (0.2 mmol) and methylenecyclopropane 2 (0.4 mmol, 2.0 equiv) under air atmosphere, DCE (2.0 mL) was added via syringe. Then, th[e reaction](#page-1-0) vesse[l](#page-2-0) was sealed with a Teflon cap. The reaction mixture was stirred vigorously at 25 °C for 24−36 h until substrate 1 disappeared (monitored by TLC). At this time, the solvent was removed under reduced pressure, and the crude product was purified by flash chromatography (the crude residue was dry loaded with silica gel, 1/ 100 to 1/50, ethyl acetate/petroleum ether) to provide product 3.

3a, a white solid, 0.096 g, 99% yield. Mp: 120−121 °C; ¹ H NMR $(400 \text{ MHz}, \text{ DMSO-}d_6): \delta 7.84 \text{ (d, } J = 6.4 \text{ Hz}, 1H), 7.76 \text{ (d, } J = 6.0 \text{ Hz},$ 1H), 7.61 (d, J = 5.6 Hz, 1H), 7.48−7.42 (m, 3H), 7.35−7.32 (m, 2H), 7.27−7.22 (m, 3H), 7.19−7.16 (m, 1H), 7.13 (t, J = 6.0 Hz, 1H), 6.83–6.80 (m, 2H), 6.66 (t, J = 6.0 Hz, 1H), 6.34 (d, J = 6.4 Hz, 2H), 6.14 (d, J = 12.8 Hz, 1H), 5.58–5.52 (m, 1H), 4.38 (s, 1H), 4.14–4.04 (m, 2H), 1.45−1.39 (m, 2H), 1.22−1.19 (m, 1H), 1.10−1.05 (m, 1H); 13C NMR (100 MHz, DMSO-d6): ^δ 168.3, 146.7, 145.6, 141.9, 140.9, 136.2, 132.8, 129.8, 129.3, 129.0, 128.7, 128.3, 128.1, 127.3, 126.8, 126.7, 126.6, 125.0, 123.1, 122.8, 120.8, 120.2, 116.9, 81.1, 64.5, 63.6, 61.3, 12.8, 9.0; IR (thin film) 3054, 2914, 2862, 1743, 1448, 1303, 1184, 969, 742, 693 cm⁻¹; HRMS (ESI) m/z calcd for $C_{33}H_{28}NO_3$ (M $+$ H)⁺ 486.2069, found 486.2053.

3b, a white solid, 0.038 g, 37% yield. Mp: 105−106 °C; ¹H NMR (400 MHz, CDCl3): δ 7.84−7.82 (m, 1H), 7.70−7.68 (m, 1H), 7.50− 7.48 (m, 1H), 7.38−7.34 (m, 3H), 7.31−7.21 (m, 3H), 7.18 (d, J = 7.2 Hz, 2H), 7.12−7.10 (m, 2H), 6.48 (d, J = 8.8 Hz, 2H), 6.36 (d, J = 8.8 Hz, 2H), 6.12 (d, J = 16.0 Hz, 1H), 5.54–5.47 (m, 1H), 4.29 (s, 1H), 4.19−4.07 (m, 2H), 3.53 (s, 3H), 1.61−1.55 (m, 1H), 1.49−1.35 (m, 2H), 0.94−0.88 (m, 1H), ¹³C NMR (100 MHz, CDCl₃): δ 168.4, 155.6, 144.7, 141.4, 141.2, 140.7, 139.2, 136.1, 133.4, 129.1, 128.7, 128.4, 128.3, 127.8, 127.1, 126.9, 126.5, 124.5, 122.2, 119.9, 119.6, 119.4, 112.7, 81.7, 64.7, 63.1, 61.1, 55.0, 12.8, 9.0; IR (thin film) 3021, 2930, 2838, 1735, 1504, 1178, 966, 832, 741 cm⁻¹; HRMS (ESI) *m/z* calcd for $C_{34}H_{30}NO_4$ $(M + H)^+$ 516.2175, found 516.2160.

3c, a white solid, 0.088 g, 76% yield. Mp: 114−115 °C; ¹H NMR (400 MHz, CDCl3): δ 7.82−7.80 (m, 1H), 7.65 (d, J = 6.4 Hz, 1H), 7.53−7.51 (m, 1H), 7.39−7.37 (m, 3H), 7.32−7.26 (m, 3H), 7.13− 7.08 (m, 3H), 6.97 (t, J = 7.2 Hz, 1H), 6.68 (d, J = 8.0 Hz, 2H), 6.51– 6.46 (m, 4H), 6.14 (d, J = 16.0 Hz, 1H), 5.54−5.47 (m, 1H), 4.27 (s,

1H), 4.20−4.08 (m, 2H), 1.60−1.56 (m, 1H), 1.52−1.46 (m, 1H), 1.43−1.37 (m, 1H), 0.97−0.91 (m, 1H); 13C NMR (100 MHz, CDCl3): δ 168.3, 157.9, 152.0, 145.0, 142.2, 141.5, 141.1, 140.7, 136.1, 133.5, 129.3, 129.2, 128.8, 128.4, 128.0, 127.9, 127.0, 126.9, 126.5, 124.5, 122.3, 122.1, 120.0, 119.4, 119.0, 118.8, 117.6, 81.6, 64.8, 63.3, 61.2, 12.7, 9.0; IR (thin film) 3030, 2940, 1738, 1488, 1230, 1158, 965, 838, 739 cm⁻¹; HRMS (ESI) m/z calcd for C₃₉H₃₂NO₄ (M + H)⁺ 578.2331, found 578.2319.

3d, a white solid, 0.079 g, 79% yield. Mp: 92–93 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.82 (d, J = 5.0 Hz, 1H), 7.66 (d, J = 6.0 Hz, 1H), 7.52−7.51 (m, 1H), 7.37−7.36 (m, 3H), 7.30−7.27 (m, 2H), 7.24− 7.22 (m, 1H), 7.18 (d, J = 7.5 Hz, 2H), 7.12−7.09 (m, 2H), 6.61 (d, J $= 8.5$ Hz, 2H), 6.39 (d, J = 8.5 Hz, 2H), 6.12 (d, J = 16.0 Hz, 1H), 5.53−5.47 (m, 1H), 4.26 (s, 1H), 4.15−4.07 (m, 2H), 2.00 (s, 3H), 1.56−1.53 (m, 1H), 1.50−1.45 (m, 1H), 1.42−1.37 (m, 1H), 0.92− 0.89 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.3, 145.2, 143.7, 141.7, 141.1, 140.7, 136.1, 133.5, 132.1, 129.1, 128.7, 128.3, 128.1, 127.9, 127.8, 127.1, 127.0, 126.5, 124.5, 122.2, 119.9, 119.3, 117.4, 81.3, 64.7, 63.1, 61.6, 20.5, 12.6, 9.1; IR (thin film) 3026, 2923, 2864, 1741, 1447, 1182, 967, 822, 737 cm⁻¹; HRMS (ESI) m/z calcd for $C_{34}H_{30}NO_3$ $(M + H)^+$ 500.2226, found 500.2208.

3e, a white solid, 0.091 g, 86% yield. Mp: 105−106 °C; ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3): \delta 7.79 \text{ (d, } J = 6.0 \text{ Hz}, 1H), 7.64 \text{ (d, } J = 6.0 \text{ Hz},$ 1H), 7.55 (d, J = 7.0 Hz, 1H), 7.40−7.35 (m, 3H), 7.31 (t, J = 7.0 Hz, 2H), 7.25−7.23 (m, 1H), 7.20 (d, J = 7.5 Hz, 2H), 7.13−7.08 (m, 2H), 6.68 (d, J = 8.0 Hz, 2H), 6.40 (d, J = 8.0 Hz, 2H), 6.14 (d, J = 16.0 Hz, 1H), 5.55−5.49 (m, 1H), 4.24 (s, 1H), 4.12−4.10 (m, 2H), 2.61−2.56 (m, 1H), 1.55−1.52 (m, 1H), 1.50−1.45 (m, 1H), 1.40− 1.35 (m, 1H), 1.00 (d, J = 7.0 Hz, 6H), 0.93−0.86 (m, 1H); 13C NMR $(125 \text{ MHz}, \text{CDCl}_3): \delta$ 168.4, 145.4, 144.0, 143.0, 141.8, 141.0, 140.7, 136.1, 133.4, 129.0, 128.7, 128.4, 127.9, 127.8, 127.0, 126.9, 126.5, 125.5, 124.5, 122.2, 119.9, 119.4, 117.2, 81.1, 64.7, 63.1, 61.8, 33.1, 23.8, 23.7, 12.6, 9.1; IR (thin film) 3022, 2949, 2866, 1733, 1448, 1164, 963, 832, 747 cm⁻¹; HRMS (ESI) m/z calcd for $C_{36}H_{34}NO_3$ (M + H)+ 528.2539, found 528.2529.

3f, a white solid, 0.092 g, 82% yield. Mp: 119−120 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.78 (d, J = 6.4 Hz, 1H), 7.59 (d, J = 7.6 Hz, 1H), 7.55 (d, J = 6.8 Hz, 1H), 7.41−7.35 (m, 3H), 7.31−7.26 (m, 3H), 7.19 (d, J = 7.2 Hz, 2H), 7.15−7.07 (m, 2H), 6.92 (d, J = 8.8 Hz, 2H), 6.34 (d, J = 8.8 Hz, 2H), 6.13 (d, J = 15.6 Hz, 1H), 5.53−5.46 (m, 1H), 4.23 (s, 1H), 4.15−4.08 (m, 2H), 1.58−1.53 (m, 1H), 1.53− 1.45 (m, 1H), 1.41−1.37 (m, 1H), 0.96−0.90 (m, 1H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3): \delta$ 168.0, 145.5, 145.0, 141.3, 140.9, 140.6, 136.0, 133.6, 130.5, 129.3, 129.0, 128.3, 128.1, 127.9, 127.2, 126.6, 126.5, 124.3, 122.0, 120.1, 119.5, 118.6, 115.3, 81.1, 64.8, 63.3, 61.5, 12.4, 9.0; IR (thin film) 3026, 2936, 2877, 1738, 1480, 1172, 963, 830, 741 cm⁻¹; HRMS (ESI) m/z calcd for C₃₃H₂₇BrNO₃ (M + H)⁺ 564.1174, found 564.1153.

3g, a white solid, 0.072 g, 73% yield. Mp: 117−118 °C; ¹ H NMR $(500 \text{ MHz}, \text{CDCl}_3): \delta 7.82 - 7.80 \text{ (m, 1H)}, 7.64 \text{ (d, } J = 7.5 \text{ Hz}, 1\text{H}),$ 7.53 (d, J = 8.0 Hz, 1H), 7.40–7.36 (m, 3H), 7.31 (t, J = 7.5 Hz, 2H), 7.25−7.23 (m, 1H), 7.18 (d, J = 7.0 Hz, 2H), 7.13−7.08 (m, 2H), 6.52−6.49 (m, 2H), 6.47−6.44 (m, 2H), 6.12 (d, J = 15.5 Hz, 1H), 5.53−5.47 (m, 1H), 4.27 (s, 1H), 4.18−4.08 (m, 2H), 1.59−1.56 (m, 1H), 1.49−1.45 (m, 1H), 1.41−1.37 (m, 1H), 0.95−0.92 (m, 1H); 13C NMR (125 MHz, CDCl₃): δ 168.2, 159.7 (d, J = 239.2 Hz), 144.8, 142.1 (d, J = 2.5 Hz), 141.2, 141.1, 140.6, 136.1, 133.5, 129.3, 128.9, 128.3, 128.0, 127.9, 127.0 (d, J = 15.5 Hz), 126.5, 124.4, 122.1, 120.0, 119.5, 119.2, 119.1, 114.3 (d, J = 22.7 Hz), 81.6, 64.8, 63.3, 61.2, 12.7, 9.0; IR (thin film) 3027, 2941, 2875, 1743, 1501, 964, 834, 742 cm⁻¹; HRMS (ESI) m/z calcd for $C_{33}H_{27}FNO_3$ (M + H)⁺ 504.1975, found 504.1957.

 3h , a white solid, 0.051 g, 62% yield. Mp: 133−134 °C; ^1H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: δ 7.72 (d, J = 7.0 Hz, 1H), 7.60 (d, J = 7.0 Hz, 1H), 7.53 (d, J = 8.5 Hz, 3H), 7.46 (d, J = 7.5 Hz, 1H), 7.43−7.40 (m, 1H), 7.37−7.34 (m, 1H), 7.32 (t, J = 7.0 Hz, 2H), 7.25−7.23 (m, 1H), 7.20 (d, J = 7.0 Hz, 2H), 7.15−7.12 (m, 1H), 7.08 (t, J = 7.0 Hz, 1H), 6.42 (d, J = 8.5 Hz, 2H), 6.16 (d, J = 16.0 Hz, 1H), 5.55–5.49 (m, 1H), 4.20 (s, 1H), 4.13−4.12 (m, 2H), 3.72 (s, 3H), 1.59−1.54 (m, 1H), 1.53−1.48 (m, 1H), 1.43−1.41 (m, 1H), 0.99−0.96 (m, 1H); 13C

NMR (125 MHz, CDCl₃): δ 167.9, 166.8, 150.6, 145.6, 141.7, 140.7, 136.1, 133.7, 129.7, 129.4, 129.1, 128.5, 128.4, 128.3, 127.9, 127.4, 126.5, 126.3, 124.1, 123.1, 122.0, 120.2, 119.6, 115.2, 80.6, 64.9, 63.6, 61.9, 51.5, 11.9, 9.1; IR (thin film) 3024, 2950, 2856, 1716, 1445, 1276, 1175, 964, 741 cm⁻¹; HRMS (ESI) m/z calcd for C₃₅H₃₀NO₅ $(M + H)^+$ 544.2124, found 544.2105.

3i, a white solid, 0.097 g, 94% yield. Mp: 100−101 °C; ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: δ 7.78 (d, J = 6.0 Hz, 1H), 7.62 (d, J = 6.5 Hz, 1H), 7.54 (d, J = 6.0 Hz, 1H), 7.39−7.37 (m, 3H), 7.31 (t, J = 7.0 Hz, 2H), 7.24−7.23 (m, 1H), 7.19 (d, J = 7.5 Hz, 2H), 7.12−7.09 (m, 2H), 6.30 (s, 1H), 6.13 (d, J = 16.0 Hz, 1H), 6.07 (s, 2H), 5.54–5.50 (m, 1H), 4.23 (s, 1H), 4.13−4.11 (m, 2H), 1.90 (s, 6H), 1.56−1.53 (m, 1H), 1.51−1.46 (m, 1H), 1.41−1.36 (m, 1H), 0.95−0.91 (m, 1H); 13C NMR (125 MHz, CDCl3): ^δ 168.4, 146.2, 145.5, 141.8, 141.0, 140.7, 136.9, 136.1, 133.4, 129.0, 128.7, 128.3, 127.9, 127.8, 127.0, 126.8, 126.5, 124.4, 124.3, 122.2, 119.8, 119.2, 115.0, 81.1, 64.7, 63.1, 61.6, 21.2, 12.6, 9.1; IR (thin film) 3026, 2950, 2861, 1737, 1449, 1165, 967, 740, 688 cm⁻¹; HRMS (ESI) m/z calcd for $C_{35}H_{32}NO_3$ (M + H)+ 514.2382, found 514.2366.

3j, a yellow solid, 0.097 g, 92% yield. Mp: 129−130 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.77 $(d, J = 7.2 \text{ Hz}, 1H)$, 7.60 $(d, J = 7.2 \text{ Hz},$ 1H), 7.57 (d, J = 7.2 Hz, 1H), 7.51−7.49 (m, 2H), 7.45−7.37 (m, 3H), 7.32−7.28 (m, 2H), 7.26−7.25 (m, 1H), 7.19 (d, J = 7.2 Hz, 1H), 7.15−7.06 (m, 2H), 6.90 (t, J = 8.0 Hz, 1H), 6.47 (d, J = 7.6 Hz, 1H), 6.16 (d, J = 16.0 Hz, 1H), 5.55−5.48 (m, 1H), 4.26 (s, 1H), 4.20−4.09 (m, 2H), 1.62−1.58 (m, 1H), 1.54−1.49 (m, 1H), 1.01− 0.95 (m, 1H), 0.89–0.83 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 167.8, 147.8, 147.6, 144.7, 141.0, 140.8, 140.7, 136.0, 133.8, 129.7, 129.3, 128.5, 128.4, 128.3, 127.9, 127.5, 126.5, 126.4, 124.2, 122.0, 121.9, 120.3, 119.7, 116.9, 111.6, 81.1, 64.9, 63.8, 61.4, 12.2, 9.1; IR (thin film) 3029, 2949, 2860, 1740, 1521, 1342, 1168, 966, 738, 691 cm⁻¹; HRMS (ESI) m/z calcd for C₃₃H₂₇N₂O₅ (M + H)⁺ 531.1920, found 531.1904.

3k, a white solid, 0.053 g, 47% yield. Mp: 38–39 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.75 (d, J = 7.2 Hz, 1H), 7.44 (d, J = 7.6 Hz, 2H), 7.38 (d, J = 7.6 Hz, 1H), 7.23–7.16 (m, 7H), 7.12 (d, J = 7.2 Hz, 2H), 7.08 (d, J = 7.6 Hz, 1H), 7.03 (t, J = 7.6 Hz, 1H), 6.90 (d, J = 7.6 Hz, 1H), 6.83 (t, J = 7.2 Hz, 1H), 6.06 (d, J = 16.0 Hz, 1H), 5.50−5.43 (m, 1H), 4.26 (s, 1H), 4.17−4.12 (m, 1H), 4.08−4.03 (m, 1H), 1.63−1.57 ¹³C NMR (100 MHz, CDCl₃): δ 167.3, 145.4, 143.3, 142.1, 141.7, 140.2, 136.0, 133.8, 132.9, 129.0, 128.8, 128.4, 127.9, 127.5, 127.1, 127.0, 126.9, 126.6, 126.5, 125.0, 123.7, 122.0, 120.6, 119.6, 119.5, 81.1, 65.1, 64.8, 58.4, 10.8, 10.3; IR (thin film) 3058, 2927, 2852, 1737, 1451, 1221, 965, 740, 692 cm⁻¹; HRMS (ESI) m/z calcd for $C_{33}H_{27}BrNO_3$ $(M + H)^+$ 564.1174, found 564.1157.

3l, a white solid, 0.110 g, 86% yield. Mp: 143−144 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.91 (s, 1H), 7.77 (s, 1H), 7.52 (d, J = 7.6 Hz, 1H), 7.35−7.21 (m, 6H), 7.15 (s, 2H), 6.89−6.85 (m, 2H), 6.75 (t, J = 7.2 Hz, 1H), 6.48 (d, J = 8.0 Hz, 2H), 6.22 (d, J = 16.0 Hz, 1H), 5.63– 5.56 (m, 1H), 4.32−4.27 (m, 1H), 4.22 (s, 1H), 4.16−4.11 (m, 1H), 1.53−1.47 (m, 2H), 1.42−1.36 (m, 1H), 0.98−0.92 (m, 1H); 13C NMR (100 MHz, CDCl₃): δ 167.7, 147.0, 145.7, 143.5, 139.0, 138.6, 136.0, 134.3, 132.5, 132.1, 130.0, 128.4, 128.1, 128.0, 127.9, 126.6, 123.4, 122.1, 121.7, 121.4, 121.2,120.7, 117.3, 80.9, 65.1, 63.3, 61.6, 12.6, 9.0; IR (thin film) 3023, 2926, 2867, 1752, 1595, 1184, 968, 814, 734 cm⁻¹; HRMS (ESI) *m/z* calcd for C₃₃H₂₆Br₂NO₃ (M + H)⁺ 642.0279, found 642.0271.

3m (dr = 1:1), a white solid, 0.105 g, 93% yield, mp: 39−40 °C. Isomer 1: ¹H NMR (400 MHz, CDCl₃): δ 7.93 (s, 1H), 7.79–7.76 (m, 2H), 7.38−7.18 (m, 8H), 7.12−7.09 (m, 2H), 6.86−6.82 (m, 4H), 6.20−6.12 (m, 1H), 5.64−5.56 (m, 1H), 4.32−4.26 (m, 2H), 4.21 (s, 1H), 1.52−1.46 (m, 2H), 1.42−1.37 (m, 1H), 0.96−0.93 (m, 1H); 13C NMR (100 MHz, CDCl₃): δ 167.9, 147.6, 146.1, 141.5, 140.1, 139.7, 136.1, 134.0, 132.3, 130.0, 129.0, 128.5, 128.0, 127.8, 127.5, 126.6, 124.5, 123.0, 122.0, 121.6, 120.8, 120.1, 117.3, 81.2, 65.0, 63.3, 61.7, 12.6, 9.1. Isomer 2: ¹H NMR (400 MHz, CDCl₃): δ 7.61−7.59 (m, 1H), 7.52−7.49 (m, 2H), 7.38−7.18 (m, 8H), 6.72−6.68 (m, 2H), 6.49−6.45 (m, 4H), 6.20−6.12 (m, 1H), 5.54−5.47 (m, 1H), 1.52− 1.46 (m, 2H), 1.42−1.37 (m, 1H), 0.96−0.91 (m, 1H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3)$: δ 167.9, 146.0, 144.9, 143.8, 140.0, 139.6, 136.0, 133.8, 131.8, 130.0, 129.4, 128.4, 127.9, 127.7, 126.9, 126.5, 124.5, 122.9, 121.9, 121.3, 120.6, 119.5, 117.1, 81.1, 64.8, 63.2, 61.6, 12.4, 9.1; IR (thin film) 3026, 2925, 2854, 1742, 1596, 1172, 965, 744, 692 cm⁻¹; HRMS (ESI) m/z calcd for C₃₃H₂₇BrNO₃ (M + H)⁺ 564.1174, found 564.1168.

3ab, a white solid, 0.075 g, 75% yield. Mp: 104−105 °C; $^1\rm H$ NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta \text{ 7.81}-7.79 \text{ (m, 1H)}, 7.65 \text{ (d, } J = 7.2 \text{ Hz}, 1 \text{ H}),$ 7.55−7.53 (m, 1H), 7.44 (d, J = 7.2 Hz, 1H), 7.37−7.35 (m, 2H), 7.31−7.27 (m, 2H), 7.21−7.12 (m, 3H), 7.09 (d, J = 7.2 Hz, 2H), 6.84−6.80 (m, 2H), 6.69 (t, J = 7.2 Hz, 1H), 7.49 (d, J = 8.0 Hz, 2H), 6.00 (s, 1H), 4.30 (s, 1H), 4.13 (d, $J = 12.0$ Hz, 1H), 3.87 (d, $J = 12.4$ Hz, 1H), 1.58−1.54 (m, 1H), 1.51−1.47 (m, 1H), 1.43−1.37 (m, 1H), 1.35 (s, 3H), 0.95−0.93 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 168.3, 146.3, 145.3, 141.7, 141.0, 140.6, 136.8, 131.8, 129.1, 128.7, 128.2, 128.0, 127.9, 127.6, 127.1, 126.8, 126.6, 124.4, 122.7, 120.0, 119.4, 117.1, 81.3, 70.5, 63.2, 61.7, 15.0, 12.6, 9.1; IR (thin film) 3057, 2920, 2867, 1733, 1448, 1182, 1027, 741, 696 cm $^{-1}$; HRMS (ESI) m/z calcd for $C_{34}H_{30}NO_3$ $(M + H)^+$ 500.2226, found 500.2207.

3ac, a white solid, 0.093 g, 96% yield. Mp: 120−121 °C; ¹ H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: δ 7.80 (d, J = 7.0 Hz, 1H), 7.63 (d, J = 7.5 Hz, 1H), 7.59 (d, J = 7.5 Hz, 1H), 7.47 (d, J = 7.0 Hz, 1H), 7.41−7.35 (m, 2H), 7.32−7.27 (m, 5H), 7.17 (t, J = 7.0 Hz, 1H), 7.12 (d, J = 7.0 Hz, 1H), 6.84 (t, J = 7.5 Hz, 2H), 6.69 (t, J = 7.0 Hz, 1H), 6.49 (d, J = 8.0 Hz, 2H), 4.44 (d, $J = 15.5$ Hz, 1H), 4.25 (s, 1H), 4.17 (d, $J = 15.5$ Hz, 1H), 1.52−1.49 (m, 2H), 1.43−1.41 (m, 1H), 0.99−0.94 (m, 1H); 13C NMR (125 MHz, CDCl₃): δ 168.0, 146.3, 145.2, 141.5, 140.9, 140.7, 131.8, 129.2, 128.8, 128.6, 128.1, 128.0, 127.6, 127.0, 126.8, 124.4, 122.7, 122.1, 120.1, 119.5, 117.1, 86.0, 82.2, 81.2, 63.2, 61.8, 52.5, 12.7, 9.1; IR (thin film) 3053, 2955, 2850, 1744, 1592, 1489, 1167, 744, 692 cm⁻¹; HRMS (ESI) m/z calcd for C₃₃H₂₆NO₃ (M + H)⁺ 484.1913, found 484.1898.

3ad, a white solid, 0.089 g, 97% yield. Mp: 93−94 °C; ¹ H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: δ 7.78 (d, J = 7.5 Hz, 1H), 7.63 (d, J = 7.5 Hz, 1H), 7.54 (d, J = 7.0 Hz, 1H), 7.41−7.33 (m, 3H), 7.20−7.09 (m, 5H), 6.83 (t, J = 7.5 Hz, 2H), 6.74 (d, J = 7.5 Hz, 2H), 6.69 (t, J = 7.0 Hz, 1H), 6.48 (d, J = 8.0 Hz, 2H), 4.60 (d, J = 13.0 Hz, 1H), 4.38 (d, J = 13.0 Hz, 1H), 4.27 (s, 1H), 1.53−1.45 (m, 2H), 1.41−1.37 (m, 1H), 0.94−0.89 (m, 1H); ¹³C NMR (125 MHz, CDCl₃): δ 168.3, 146.3, 145.2, 141.6, 140.9, 140.6, 134.8, 129.1, 128.8, 128.1, 128.0, 127.8, 127.7, 127.6, 127.0, 126.8, 124.4, 122.7, 120.0, 119.5, 117.2, 81.3, 66.2, 63.2, 61.6, 12.6, 9.1; IR (thin film) 3031, 2946, 2890, 1738, 1447, 1168, 1015, 742, 692 cm⁻¹; HRMS (ESI) m/z calcd for C₃₁H₂₆NO₃ $(M + H)^+$ 460.1913, found 460.1895.

3ae, a white solid, 0.069 g, 81% yield. Mp: 87−88 °C; ¹ H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.78 (d, J = 6.8 Hz, 1H), 7.64–7.59 (m, 2H), 7.50 (d, J = 7.6 Hz, 1H), 7.42−7.34 (m, 2H), 7.22 (d, J = 7.6 Hz, 1H), 7.15 (t, J = 7.6 Hz, 1H), 6.84–6.80 (m, 2H), 6.69 (d, J = 7.2 Hz, 1H), 6.48 (d, J = 8.0 Hz, 2H), 4.23 (s, 1H), 3.49–3.46 (m, 2H), 1.56–1.52 (m, 1H), 1.48−1.37 (m, 2H), 0.97−0.87 (m, 3H), 0.85−0.78 (m, 2H), 0.66 (t, J = 6.8 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 168.5, 146.4, 145.3, 141.8, 141.0, 140.7, 129.1, 128.7, 128.0, 127.6, 127.0, 126.9, 124.4, 122.6, 119.9, 119.3, 117.1, 81.2, 64.1, 63.2, 61.5, 29.8, 18.6, 13.5, 12.4, 9.1; IR (thin film) 3042, 2957, 2871, 1730, 1450, 1187, 745, 690 cm⁻¹; HRMS (ESI) m/z calcd for C₂₈H₂₈NO₃ (M + H)⁺ 426.2069, found 426.2050.

3af, a white solid, 0.071 g, 90% yield. Mp: 116−117 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.78 (d, J = 6.8 Hz, 1H), 7.63–7.61 (m, 2H), 7.51 (d, J = 7.6 Hz, 1H), 7.42−7.34 (m, 2H), 7.22 (d, J = 7.6 Hz, 1H), 7.14 (t, J = 7.2 Hz, 1H), 6.84–6.80 (m, 2H), 6.69 (d, J = 7.2 Hz, 1H) 6.48 (d, J = 8.0 Hz, 2H), 4.19 (s, 1H), 3.60−3.48 (m, 2H), 1.56−1.53 $(m, 1H)$, 1.51−1.35 $(m, 2H)$, 0.95−0.88 $(m, 1H)$, 0.59 $(t, J = 7.2$ Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 168.4, 146.4, 145.5, 141.9, 140.9, 140.7, 129.1, 128.7, 128.0, 127.6, 127.0, 126.8, 124.4, 122.6, 119.9, 119.3, 117.0, 81.2, 63.2, 61.6, 60.2, 13.2, 12.4, 9.0; IR (thin film) 3065, 2985, 1733, 1449, 1193, 741, 696 cm[−]¹ ; HRMS (ESI) m/z calcd for $C_{26}H_{24}NO_3$ $(M + H)^+$ 398.1756, found 398.1741.

3ag, a white solid, 0.060 g, 71% yield. Mp: 107−108 °C; ¹ H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.67 $(d, J = 7.2 \text{ Hz}, 1H)$, 7.63 $(d, J = 7.6 \text{ Hz},$ 1H), 7.52 (d, J = 6.8 Hz, 1H), 7.44 (d, J = 7.6 Hz, 1H), 7.33−7.25 (m, 2H), 7.17−7.13 (m, 1H), 7.09 (d, J = 7.6 Hz, 1H), 6.76−6.72 (m, 2H), 6.62 (t, J = 7.2 Hz, 1H), 6.42 (d, J = 8.0 Hz, 2H), 4.17 (s, 1H), 1.68−1.66 (m, 1H), 1.40−1.30 (m, 2H), 0.85−0.78 (m, 1H), 0.70 (s, 9H); ¹³C NMR (100 MHz, CDCl₃): δ 167.0, 146.4, 145.2, 142.2, 141.3, 140.9, 128.9, 128.6, 127.8, 127.5, 127.2, 127.1, 124.4, 122.6, 119.7, 119.2, 117.3, 81.2, 80.7, 63.0, 60.9, 27.0, 11.9, 8.9; IR (thin film) 3054, 2974, 1730, 1451, 1151, 844, 746, 694 cm⁻¹; HRMS (ESI) m/z calcd for $C_{28}H_{28}NO_3$ $(M + H)^+$ 426.2069, found 426.2052.

3ah (dr = 3.5/1), a white solid, 0.058 g, 58% yield, mp: 86−87 °C. Major isomer: ¹H NMR (400 MHz, CDCl₃): δ 7.65 (d, J = 7.2 Hz, 1H), 7.61 (d, J = 7.2 Hz, 1H), 7.53 (d, J = 7.2 Hz, 1H), 7.42 (d, J = 7.2 Hz, 1H), 7.31−7.24 (m, 2H), 7.18−7.13 (m, 1H), 7.09−7.05 (m, 1H), $6.77-6.73$ (m, 2H), $6.63-6.59$ (m, 1H), 6.43 (d, J = 8.0 Hz, 2H), 4.24 (s, 1H), 4.09−4.02 (m, 1H), 1.62−1.58 (m, 2H), 1.39−1.30 (m, 6H), 1.12−1.07 (m, 1H), 0.81−0.78 (m, 2H), 0.70−0.67 (m, 2H), 0.57− 0.51 (m, 6H), 0.40 (d, J = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 167.6, 146.4, 145.1, 142.3, 141.4, 141.1, 129.0, 128.6, 127.8, 127.5, 127.2, 126.8, 124.4, 122.7, 119.7, 119.4, 117.4, 80.9, 74.3, 63.2, 60.4, 46.1, 38.7, 33.8, 30.7, 25.9, 23.2, 21.7, 20.4, 16.1, 11.7, 9.0. Minor isomer: ¹H NMR (400 MHz, CDCl₃): δ 7.70 (d, J = 7.2 Hz, 1H), 7.69 $(d, J = 7.2 \text{ Hz}, 1\text{H}), 7.49 \ (d, J = 7.2 \text{ Hz}, 1\text{H}), 7.37 \ (d, J = 7.2 \text{ Hz}, 1\text{H}),$ 7.31−7.24 (m, 2H), 7.18−7.13 (m, 1H), 7.09−7.05 (m, 1H), 6.77− 6.73 (m, 2H), 6.63−6.59 (m, 1H), 6.39 (d, J = 8.0 Hz, 2H), 4.33 (s, 1H), 4.09−4.02 (m, 1H), 1.62−1.58 (m, 2H), 1.43−1.40 (m, 6H), 1.10−1.07 (m, 1H), 0.81−0.78 (m, 2H), 0.68−0.67 (m, 2H), 0.57− 0.51 (m, 6H), 0.43 (d, J = 7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 167.7, 146.0, 144.0, 142.3, 141.5, 140.4, 129.2, 128.7, 127.7, 127.5, 127.4, 127.0, 124.6, 123.0, 119.8, 119.3, 117.8, 81.7, 74.6, 63.7, 61.1, 46.1, 40.4, 33.9, 31.1, 26.8, 22.5, 21.8, 20.8, 15.3, 12.5, 9.5; IR (thin film) 3062, 2953, 2867, 1735, 1595, 1454, 1184, 747, 693 cm⁻¹; HRMS (ESI) m/z calcd for $C_{34}H_{38}NO_3$ $(M + H)^+$ 508.2852, found 508.2836.

3ai, a white solid, 0.063 g, 73% yield. Mp: 120−121 °C; ¹ H NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta 8.01 \text{ (d, } J = 7.6 \text{ Hz}, 1H), 7.47-7.45 \text{ (m, } 1H),$ 7.41−7.39 (m, 3H), 7.21 (t, J = 7.6 Hz, 1H), 7.12−7.10 (m, 1H), 7.00−6.98 (m, 2H), 6.96−6.90 (m, 4H), 6.83−6.80 (m, 2H), 6.67− 6.64 (m, 1H), 6.48 (d, J = 8.4 Hz, 2H), 4.92 (s, 1H), 1.69–1.66 (m, 1H), 1.43−1.37 (m, 2H), 0.94−0.88 (m, 1H); 13C NMR (100 MHz, CDCl₃): δ 196.0, 146.6, 146.2, 141.1, 141.0, 140.3, 137.1, 132.3, 129.2, 128.4, 128.3, 128.1, 127.7, 127.5, 127.4, 127.1, 124.5, 122.3, 120.0, 119.0, 116.7, 81.0, 64.5, 63.7, 13.5, 9.3; IR (thin film) 3066, 2957, 2857, 1727, 1680, 1597, 1213, 746, 692 cm[−]¹ ; HRMS (ESI) m/z calcd for $C_{30}H_{24}NO_2$ $(M + H)^+$ 430.1807, found 430.1792.

3aj, as a white solid, 0.065 g, 92% yield. Mp: 149−150 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.74 $(d, J = 7.6 \text{ Hz}, 1H)$, 7.70 $(d, J = 7.2 \text{ Hz},$ 1H), 7.61 (d, J = 7.2 Hz, 1H), 7.56 (d, J = 7.2 Hz, 1H), 7.41−7.32 (m, 2H), 7.29−7.25 (m, 1H), 7.21−7.17 (m, 1H), 6.84−6.80 (m, 2H), 6.69 (t, J = 7.2 Hz, 1H), 6.49 (d, J = 8.0 Hz, 2H), 3.71−3.68 (m, 1H), 3.32−3.28 (m, 1H), 3.15−3.11 (m, 1H), 1.41−1.36 (m, 1H), 1.21− 1.12 (m, 2H), 0.93–0.87 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 146.6, 144.9, 141.7, 140.6, 140.5, 129.0, 128.6, 128.0, 127.6, 127.1, 126.8, 124.8, 122.5, 120.0, 119.9, 117.2, 80.9, 65.1, 61.0, 57.0, 10.7, 7.2; IR (thin film) 3660, 3030, 2927, 2865, 1592, 1217, 740, 691 cm⁻¹; HRMS (ESI) m/z calcd for $C_{24}H_{22}NO_2$ (M + H)⁺ 356.1651, found 356.1633.

Direct Thermal Rearrangement of 3a to 4a' (Scheme 2-1). A round-bottle flask was charged with 3a (0.2 mmol) and toluene (2 mL). Then, the reaction mixture was stirred vigorously at 100 °C for 36 h until the substrate 3a disappeared (monitored [by](#page-2-0) [TLC\).](#page-2-0) [A](#page-2-0)t this time, the solvent was removed under reduced pressure, and the crude product was purified by flash chromatography (the crude residue was dry loaded with silica gel, 1/100 to 1/10, ethyl acetate/petroleum ether) to provide 1a (0.020 g, 35% yield), N-(9H-fluoren-9 ylidene)aniline (0.010 g, 20% yield) and 4a′ (0.010 g, 10% yield).

Cinnamyl4′-hydroxy-1′-phenyl-5′,6′-dihydro-1′H-spiro[fluorene-9,2'-pyridine]-3'-carboxylate (4a'). A white solid, mp: 126–127 °C; ¹H NMR (400 MHz, CDCl₃): δ 12.90 (s, 1H), 7.48 (d, J = 7.2 Hz, 2H), 7.30−7.04 (m, 11H), 6.67−6.59 (m, 3H), 6.20 (d, J = 7.2 Hz, 2H), 5.78 (d, J = 16.0 Hz, 1H), 5.19−5.12 (m, 1H), 4.05 (d, J = 7.0 Hz, 2H), 3.51 (t, $J = 5.2$ Hz, 2H), 2.79 (t, $J = 5.2$ Hz, 2H); ¹³C NMR (100 MHz, CDCl3): δ 174.0, 170.8, 149.2, 148.2, 141.2, 136.3, 132.8, 128.3, 127.8, 127.6, 127.4, 126.7, 126.4, 124.2, 124.1, 122.1, 119.1, 100.9, 70.4, 63.8, 45.4, 31.2; IR (thin film) 3442, 3040, 2954, 2847, 1725, 1640, 1247, 1072, 740, 694 cm⁻¹; HRMS (ESI) m/z calcd for $C_{33}H_{28}NO_3$ (M + H)⁺ 486.2069, found 486.2059.

General Procedure B for Preparation of Product 4 (Table 4). In a round-bottle flask charged with 3 (0.2 mmol) and THF (2 mL) was added LiAl H_4 (0.4 mmol, 2.0 equiv). Then, the reaction mixture was stirred vigorously at 25 °C for 1 h until the substrate 3 [disapp](#page-2-0)eared (monitored by TLC). At this time, the reaction was diluted with H_2O (10 mL) and exacted with Et₂O (3 \times 10 mL). The combined organic layers were washed with brine $(1 \times 10 \text{ mL})$, dried over Na₂SO₄, and filtered. The solvent was removed under reduced pressure, and the crude product was used directly in the next step.

A Teflon-sealed reaction flask was charged with the above crude mixture and MeCN (2 mL). Then, the reaction mixture was stirred vigorously at 120 °C for 18−24 h until the substrate disappeared (monitored by TLC). At this time, the solvent was removed under reduced pressure, and the crude product was purified by flash chromatography (the crude residue was dry-loaded with silica gel, 1/ 20 to 1/4 ethyl acetate/petroleum ether) to provide product 4.

3′-(Hydroxymethyl)-1′-phenylspiro[fluorene-9,2[°]-piperidin]-4′-
one (4a). A white solid, 0.050 g, 70% yield. Mp: 116–117 °C; ¹H NMR (400 MHz, DMSO- d_6): δ 7.74 (d, J = 7.2 Hz, 2H), 7.63–7.61 $(m, 1H)$, 7.40 $(t, J = 7.2$ Hz, 1H), 7.30–7.19 $(m, 4H)$, 6.77 $(t, J = 7.6)$ Hz, 2H), 6.61−6.58 (m, 1H), 6.35 (d, J = 7.6 Hz, 2H), 4.29−4.27 (m, 1H), 4.13 (brs, 1H), 3.85−3.82 (m, 1H), 3.67 (d, J = 7.2 Hz, 1H), 3.10−3.01 (m, 2H), 2.90−2.82 (m, 1H), 2.10 (d, J = 5.6 Hz, 1H); ¹³C NMR (100 MHz, DMSO-d₆): δ 208.9, 149.4, 147.2, 146.3, 140.1, 139.9, 129.1, 128.8, 128.2, 128.1, 127.8, 125.0, 124.9, 122.5, 122.4, 120.9, 120.2, 73.7, 57.9, 55.7, 47.3, 42.8; IR (thin film) 3657, 3030, 2927, 2865, 1739, 1592, 1217, 740, 691 cm^{−1}; HRMS (ESI) *m/z* calcd for $C_{24}H_{22}NO_2$ (M + H)⁺ 356.1651, found 356.1635.

3′-(Hydroxymethyl)-1′-(4-phenoxyphenyl)spiro[fluorene-9,2′-piperidin]-4′-one (4c). A white solid, 0.077 g, 87% yield. Mp: 38–39 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.61 (d, J = 6.5 Hz, 1H), 7.54 (d, J = 7.0 Hz, 1H), 7.39−7.34 (m, 3H), 7.28−7.20 (m, 5H), 7.01 (t, J = 7.0 Hz, 1H), 6.71 (d, J = 7.5 Hz, 2H), 6.44 (t, J = 8.0 Hz, 2H), 6.41 (d, J = 8.0 Hz, 2H), 4.02−3.98 (m, 1H), 3.69−3.65 (m, 2H), 3.37−3.33 (m, 1H), 3.20−3.17 (m, 1H), 2.88 (d, J = 15.0 Hz, 1H), 2.59−2.58 (m, 1H), 2.27 (brs, 1H); ¹³C NMR (125 MHz, CDCl₃): δ 211.8, 157.5, 153.5, 145.4, 143.8, 143.7, 140.7, 139.7, 129.4, 128.9, 128.7, 127.7, 127.3, 127.2, 124.9, 124.6, 122.7, 120.5, 119.5, 118.3, 118.0, 73.7, 58.5, 58.3, 49.4, 42.4; IR (thin film) 3439, 3034, 2930, 2851, 1708, 1621, 1231, 869, 741, 698 cm⁻¹; HRMS (ESI) m/z calcd for $C_{30}H_{26}NO_3$ (M + H)+ 448.1913, found 448.1905.

3′-(Hydroxymethyl)-1′-p-tolylspiro[fluorene-9,2′-piperidin]-4′ one (4d). A white solid, 0.055 g, 74% yield. Mp: 40−41 °C; ¹ H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: δ 7.61 (d, J = 7.5 Hz, 1H), 7.54 (d, J = 7.5 Hz, 1H), 7.38−7.34 (m, 3H), 7.29−7.18 (m, 3H), 6.59 (d, J = 7.5 Hz, 2H), 6.35 (d, J = 8.0 Hz, 2H), 4.05−3.99 (m, 1H), 3.72−3.68 (m, 1H), 3.64 (dd, J = 9.0 Hz, 3.0 Hz, 1H), 3.32−3.28 (m, 1H), 3.19−3.14 $(m, 1H)$, 2.87 (dt, J = 15.0 Hz, 4.0 Hz, 1H), 2.56–2.54 $(m, 1H)$, 2.25 (brs, 1H), 2.03 (s, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 211.9, 145.7, 145.6, 144.3, 140.6, 139.6, 133.9, 128.7, 128.6, 128.3, 127.7, 127.2, 125.5, 124.9, 124.6, 120.4, 119.5, 73.5, 58.7, 58.3, 49.4, 42.5, 20.6; IR (thin film) 3440, 3060, 2929, 2851, 1709, 1622, 1215, 820, 739, 694 cm⁻¹; HRMS (ESI) m/z calcd for C₂₅H₂₃NO₂Na (M + Na)⁺ 392.1626, found 392.1609.

3′-(Hydroxymethyl)-1′-(4-isopropylphenyl)spiro[fluorene-9,2′-piperidin]-4′-one (4e). A white solid, 0.061 g, 77% yield. Mp: 125−126 $^{\circ}$ C; ¹H NMR (500 MHz, CDCl₃): δ 7.52 (d, J = 7.0 Hz, 1H), 7.48 (d, J = 7.5 Hz, 1H), 7.32−7.25 (m, 2H), 7.21−7.19 (m, 2H), 7.16−7.10 $(m, 2H)$, 6.57 (d, J = 8.0 Hz, 2H), 6.29 (d, J = 8.5 Hz, 2H), 3.98–3.93 (m, 1H), 3.72−3.68 (m, 1H), 3.57−3.55 (m, 1H), 3.24−3.20 (m, 1H), 3.12−3.06 (m, 1H), 2.82 (dt, J = 15.0 Hz, 4.0 Hz, 1H), 2.55−2.49 (m, 2H), 2.17 (brs, 1H), 0.94 (d, $J = 6.5$ Hz, 6H); ¹³C NMR (100 MHz, CDCl3): δ 211.9, 145.9, 145.7, 144.5, 140.5, 139.6, 128.7, 128.5, 127.7, 127.2, 125.6, 125.0, 124.9, 124.8, 124.5, 120.4, 119.4, 73.4, 58.8, 58.2, 49.2, 42.5, 33.1, 23.7, 23.6; IR (thin film) 3446, 3030, 2963, 2875,

1711, 1616, 1047, 739, 668 cm⁻¹; HRMS (ESI) m/z calcd for $C_{27}H_{28}NO_2$ (M + H)⁺ 398.2120, found 398.2105.

1′-(4-Bromophenyl)-3′-(hydroxymethyl)spiro[fluorene-9,2′-piperidin]-4′-one (**4f**). A white solid, 0.041 g, 47% yield. Mp: 42–43 °C; ¹H NMR (500 MHz, CDCl₃): δ 7.58–7.55 (m, 2H), 7.50–7.33 (m, 2H), 7.29−7.25 (m, 1H), 7.22−7.18 (m, 2H), 6.81−6.78 (m, 2H), 6.76− 6.74 (m, 1H), 6.45 (d, J = 7.5 Hz, 2H), 4.07−4.02 (m, 1H), 3.84−3.79 $(m, 1H)$, 3.67 (dd, J = 8.0 Hz, 2.5 Hz, 1H), 3.32–3.28 $(m, 1H)$, 3.32– 3.13 (m, 1H), 2.92 (dt, J = 15.5 Hz, 4.5 Hz, 1H), 2.56−2.51 (m, 1H), 2.23 (brs, 1H); ¹³C NMR (125 MHz, CDCl₃): δ 211.7, 148.3, 145.6, 144.5, 140.4, 139.7, 128.8, 128.6, 127.7, 127.3, 125.0, 124.9, 124.8, 124.5, 124.1, 120.5, 119.5, 73.4, 58.7, 58.2, 48.8, 42.4; IR (thin film) 3415, 3059, 2927, 2877, 1715, 1593, 1486, 1035, 741, 697 cm⁻¹; HRMS (ESI) m/z calcd for C₂₄H₁₉BrNO (M – OH)⁺ 416.0650, found 416.0630.

1′-(4-Fluorophenyl)-3′-(hydroxymethyl)spiro[fluorene-9,2′-piperidin]-4′-one (4g). A white solid, 0.062 g, 82% yield. Mp: 139–140 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.62 (d, J = 7.2 Hz, 1H), 7.54 (d, J = 7.2 Hz, 1H), 7.40−7.34 (m, 3H), 7.30−7.18 (m, 3H), 6.49−6.45 (m, 2H), 6.42−6.38 (m, 2H), 4.01−3.94 (m, 1H), 3.65−3.59 (m, 2H), 3.36−3.31 (m, 1H), 3.22−3.14 (m, 1H), 2.87 (dt, J = 14.8 Hz, 3.6 Hz, 1H), 2.59−2.54 (m, 1H), 2.25 (s, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 211.6, 160.9 (d, J = 242.8 Hz), 145.2, 144.0, 143.6, 140.7, 139.6, 128.9 (d, J = 12.5 Hz), 127.8, 127.7, 127.6, 127.2, 124.9, 124.5, 120.5, 119.6, 114.4 (d, $J = 21.9$ Hz), 73.6, 58.5, 58.2, 49.6, 42.4; IR (thin film) 3474, 3057, 2930, 2850, 1708, 1617, 1211, 837, 739 cm⁻¹; HRMS (ESI) m/z calcd for $C_{24}H_{21}FNO_2$ $(M + H)^+$ 374.1556, found 374.1544.

3′-(Hydroxymethyl)-1′-(4-(hydroxymethyl)phenyl)spiro[fluorene-9,2′-piperidin]-4′-one (4h). A white solid, 0.056 g, 73% yield. Mp: 47−48 °C; ¹ H NMR (500 MHz, CDCl3): δ 7.58−7.55 (m, 2H), 7.41−7.37 (m, 2H), 7.34−7.32 (m, 1H), 7.29−7.26 (m, 1H), 7.22− 7.18 (m, 2H), 6.80 (d, $J = 8.0$ Hz, 2H), 6.44 (d, $J = 8.0$ Hz, 2H), 4.37 (s, 2H), 4.05−4.02 (m, 1H), 3.82−3.80 (m, 1H), 3.75−3.74 (m, 1H), 3.66 (dd, J = 9.0 Hz, 3.0 Hz, 1H), 3.30−3.26 (m, 1H), 3.17−3.14 (m, 1H), 2.92 (dt, J = 15.5, Hz, 4.5 Hz, 1H), 2.54−2.53 (m, 1H), 2.22 (brs 1H); ¹³C NMR (100 MHz, CDCl₃): δ 211.6, 147.9, 145.6, 144.4, 140.4, 139.6, 136.3, 128.8, 128.7, 127.8, 127.4, 126.6, 124.9, 124.8, 124.4, 120.5, 119.6, 73.4, 64.7, 58.7, 58.2, 48.8, 42.4; IR (thin film) 3419, 3054, 2926, 2862, 1712, 1611, 1509, 1261, 1028, 805, 739 cm⁻¹; HRMS (ESI) m/z calcd for $C_{25}H_{24}NO_3$ (M + H)⁺ 386.1756, found 386.1739.

1′-(3,5-Dimethylphenyl)-3′-(hydroxymethyl)spiro[fluorene-9,2′ piperidin]-4′-one (4i). A white solid, 0.049 g, 63% yield. Mp: 154−155 $^{\circ}$ C; ¹H NMR (400 MHz, CDCl₃): δ 7.58 (d, J = 7.2 Hz, 1H), 7.54 (d, J = 7.6 Hz, 1H), 7.40−7.28 (m, 3H), 7.25−7.21 (m, 1H), 7.20−7.17 (m, 2H), 6.38 (s, 1H), 6.02 (s, 2H), 4.03−3.98 (m, 1H), 3.79−3.75 (m, 1H), 3.65−3.62 (m, 1H), 3.35−3.30 (m, 1H), 3.16−3.12 (m, 1H), 2.90−2.85 (m, 1H), 2.59 (d, J = 10.4 Hz, 1H), 2.23 (brs, 1H), 1.88 (s, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 211.9, 148.1, 145.7, 144.5, 140.5, 139.7, 136.9, 128.7, 128.6, 127.6, 127.2, 125.6, 124.8, 124.5, 122.8, 120.2, 119.4, 73.4, 58.5, 58.3, 48.8, 42.5, 20.9; IR (thin film) 3432, 3045, 2928, 2854, 1711, 1614, 1045, 744 cm[−]¹ ; HRMS (ESI) m/ z calcd for $C_{26}H_{26}NO_2 (M + H)^+$ 384.1964, found 384.1953.

3′-(Hydroxymethyl)-1′-(3-nitrophenyl)spiro[fluorene-9,2′-piperidin]-4'-one (4j). A yellow liquid, 0.040 g, 50% yield. ¹H NMR (500 MHz, CDCl₃): δ 7.63 (d, J = 8.0 Hz, 1H), 7.56 (d, J = 8.0 Hz, 1H), 7.52 (d, J = 7.5 Hz, 1H), 7.51 (d, J = 7.0 Hz, 1H), 7.49−7.47 (m, 1H), 7.44−7.41 (m, 3H), 7.33−7.22 (m, 2H), 6.92 (t, J = 7.5 Hz, 1H), 6.61 $(d, J = 7.0 \text{ Hz}, 1H)$, 4.09–3.03 (m, 2H), 3.71–3.69 (m, 1H), 3.29– 3.26 (m, 1H), 3.17−3.13 (m, 1H), 3.03−2.98 (m, 1H), 2.59−2.55 (m, 1H), 2.17 (s, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 210.5, 149.3, 147.8, 144.9, 144.0, 139.9, 139.7, 129.4, 129.2, 128.8, 128.4, 128.3, 127.9, 124.4, 123.9, 120.8, 120.0, 117.6, 117.5, 73.1, 58.1, 58.0, 47.2, 41.9; IR (thin film) 3430, 3031, 2972, 2860, 1708, 1619, 1048, 738 cm⁻¹; HRMS (ESI) m/z calcd for C₂₄H₂₁N₂O₄ (M + H)⁺ 401.1501, found 401.1493.

1′-(2-Bromophenyl)-3′-(hydroxymethyl)spiro[fluorene-9,2′-piper*idin]-4′-one (4k)*. A white solid, 0.049 g, 56% yield. Mp: 69–70 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.51–7.47 (m, 2H), 7.33–7.26 (m, 2H), 7.22−7.16 (m, 1H), 7.14−7.10 (m, 2H), 6.74−6.66 (m, 3H), 6.38 (d, J = 7.2 Hz, 2H), 4.01–3.94 (m, 1H), 3.78–3.72 (m, 1H), 3.60 (dd, J = 8.8 Hz, 3.2 Hz, 1H), 3.25−3.20 (m, 1H), 3.13−3.05 (m, 1H), 2.85−2.79 (m, 1H), 2.51 (d, J = 10.4 Hz, 1H), 2.14 (brs, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 211.7, 148.3, 145.6, 144.5, 140.4, 139.7, 128.8, 128.6, 127.7, 127.3, 125.0, 124.8, 124.5, 124.1, 120.5, 119.5, 73.4, 58.7, 58.2, 48.8, 42.4; IR (thin film) 3407, 3057, 2960, 2858, 1710, 1591, 1449, 1262, 1094, 804, 742 cm⁻¹; HRMS (ESI) m/z calcd for $C_{24}H_{21}BrNO_2$ $(M + H)^+$ 434.0756, found 434.0730.

2,7-Dibromo-3′-(hydroxymethyl)-1′-phenylspiro[fluorene-9,2′-piperidin]-4′-one (4l). A white solid, 0.066 g, 65% yield. Mp: 42–43 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.73 (s, 1H), 7.53 (d, J = 8.0 Hz, 1H), 7.46 (s, 1H), 7.37 (d, J = 8.0 Hz, 1H), 7.32 (d, J = 7.6 Hz, 1H), 7.21 (d, J = 8.0 Hz, 1H), 6.87−6.78 (m, 3H), 6.46 (d, J = 7.6 Hz, 2H), 3.99−3.93 (m, 1H), 3.83−3.77 (m, 1H), 3.59−3.57 (m, 1H), 3.36− 3.31 (m, 1H), 3.17–3.10 (m, 1H), 2.93 (dt, J = 15.6 Hz, 4.4 Hz, 1H), 2.58 -2.55 (m, 1H) 2.28 (brs, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 210.6, 147.6, 147.5, 146.4, 138.4, 137.6, 132.2, 132.0, 128.1, 128.0, 127.9, 125.2, 124.7, 121.9, 121.8, 121.7, 120.9, 73.4, 58.0, 57.9, 48.8, 42.3; IR (thin film) 3478, 3038, 2977, 2873, 1728, 1614, 1131, 880, 652 cm⁻¹; HRMS (ESI) m/z calcd for C₂₄H₂₀Br₂NO₂ (M + H)⁺ 511.9861, found 511.9866.

2-Bromo-3′-(hydroxymethyl)-1′-phenylspiro[fluorene-9,2′-piperidin]-4′-one (4m). Dr = $1/1$, a white solid, 0.040 g, 46% yield, mp: 37– 38 °C. Isomer 1: ¹H NMR (400 MHz, CDCl₃): δ 7.74 (s, 1H), 7.57− 7.46 (m, 2H), 7.36−7.19 (m, 4H), 6.82−6.73 (m, 3H), 6.44 (d, J = 7.6 Hz, 2H), 4.40−3.97 (m, 1H), 3.84−3.77 (m, 1H), 3.66−3.59 (m, 1H), 3.36−3.27 (m, 1H), 3.16−3.09 (m, 1H), 2.91−2.86 (m, 1H), 2.57− 2.54 (m, 1H), 2.35 (brs, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 211.9, 148.3, 147.7, 145.6, 144.5, 140.5, 139.8, 132.1, 129.3, 128.8, 127.8, 125.2, 124.5, 121.7, 121.2, 120.5, 119.6, 73.4, 58.6, 58.1, 48.8, 42.4. Isomer 2: ¹H NMR (400 MHz, CDCl₃): δ 7.57−7.46 (m, 3H), 7.36− 7.19 (m, 4H), 6.82−6.73 (m, 3H), 6.44 (d, J = 7.6 Hz, 2H), 4.40−3.97 (m, 1H), 3.84−3.77 (m, 1H), 3.66−3.59 (m, 1H), 3.36−3.27 (m, 1H), 3.16−3.09 (m, 1H), 2.91−2.86 (m, 1H), 2.57−2.54 (m, 1H), 2.35 (brs, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 211.1, 148.0, 146.8, 145.4, 144.0, 139.4, 138.6, 131.7, 129.0, 128.6, 127.3, 124.8, 124.0, 121.4, 120.8, 120.4, 119.3, 73.3, 58.3, 58.0, 48.6, 42.3; IR (thin film) 3406, 3013, 2932, 2867, 1719, 1631, 1250, 877, 655 cm⁻¹; HRMS (ESI) *m/z* calcd for $C_{24}H_{21}BrNO_2$ (M + H)⁺ 434.0756, found 434.0751.
3'-(Benzyloxymethyl)-1'-phenylspiro[fluorene-9,2'-piperidin]-4'-

3′-(Benzyloxymethyl)-1′-phenylspiro[fluorene-9,2′-piperidin]-4′-
one (4n). A white solid, 0.057 g, 64% yield. Mp: 94−95 °C; ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$: δ 7.63 $(d, J = 7.2 \text{ Hz}, 1H)$, 7.51 $(d, J = 7.6 \text{ Hz},$ 1H), 7.37−7.32 (m, 3H), 7.25−7.16 (m, 6H), 7.00 (d, J = 6.4 Hz, 2H), 6.79−6.70 (m, 3H), 6.42 (d, J = 7.6 Hz, 2H), 4.21 (d, J = 12.0 Hz, 1H), 4.03 (d, J = 11.6 Hz, 1H), 3.97−3.94 (m, 1H), 3.89−3.83 $(m, 1H)$, 3.78 (d, J = 6.0 Hz, 1H), 3.48 (t, J = 8.8 Hz, 1H), 3.18–3.11 (m, 1H), 2.92 (dt, J = 14.8 Hz, 4.8 Hz, 1H), 2.39 (d, J = 9.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 208.0, 148.5, 146.2, 144.5, 140.6, 139.8, 138.1, 128.7, 128.4, 128.0, 127.7, 127.6, 127.4, 127.3, 127.2, 125.1, 125.0, 124.6, 123.9, 120.2, 119.4, 74.1, 72.9, 64.7, 56.8, 49.1, 42.7; IR (thin film) 3060, 2925, 2859, 1713, 1595, 1489, 1100, 740, 698 cm⁻¹; HRMS (ESI) m/z calcd for $C_{31}H_{28}NO_2$ $(M + H)^+$ 446.2120, found 446.2114.

3′-(Allyloxymethyl)-1′-phenylspiro[fluorene-9,2′-piperidin]-4′ one (40). A colorless liquid, 0.054 g, 68% yield. ¹H NMR (400 MHz, CDCl₃): δ 7.57 (d, J = 6.8 Hz, 1H), 7.46 (d, J = 7.6 Hz, 1H), 7.32– 7.24 (m, 2H), 7.20−7.12 (m, 3H), 6.72−6.65 (m, 3H), 6.35 (d, J = 7.2 Hz, 2H), 5.56−5.46 (m, 1H), 4.88−4.84 (m, 2H), 3.91−3.88 (m, 1H), 3.84−3.78 (m, 1H), 3.66 (d, J = 7.6 Hz, 1H), 3.58 (dd, J = 12.8 Hz, 5.6 Hz, 1H), 3.46 (dd, J = 12.8 Hz, 5.6 Hz, 1H), 3.31−3.26 (m, 1H), 3.11−3.04 (m, 1H), 2.84−2.78 (m, 1H), 2.22−2.20 (m, 1H); 13C NMR (100 MHz, CDCl₃): δ 208.1, 148.5, 146.2, 144.5, 140.6, 139.8, 134.4, 128.7, 128.4, 127.7, 127.6, 127.3, 125.1, 125.0, 124.6, 123.9, 120.2, 119.3, 116.5, 74.0, 71.8, 64.3, 56.8, 49.1, 42.7; IR (thin film) 3062, 2923, 2879, 1711, 1594, 1483, 1105, 741, 693 cm⁻¹; HRMS (ESI) m/z calcd for $C_{27}H_{26}NO_2$ (M + H)⁺ 396.1964, found 396.1953.

Enantioselective Synthesis of 3aj and 4a (Scheme 3). Chiral 3aj was prepared by general procedure B. 3ah (0.030 g, 0.06 mmol) was stirred for 1 h at 25 °C. Purification us[ing mediu](#page-3-0)m pressure chromatography (1/4 ethyl acetate/petroleum ether) afforded chiral 3aj as a white solid $(0.021 \text{ g}, 86\%)$. Ee = 35%, conditions: AD-H; hexane/i-PrOH = $85/15$; flow rate 1.0 mL/min; λ = 254 nm; t $(\text{minor}) = 36.4 \text{ min};$ t $(\text{major}) = 43.2 \text{ min}.$

Chiral 4a was prepared by general procedure B. Chiral 3aj (0.021 g, 0.06 mmol) was stirred for 12 h at 120 °C. Purification using medium pressure chromatography (1/4 ethyl acetate/petroleum ether) afforded chiral 4a as a white solid (0.016 g, 89%). The characteristic data was matched with that above. Ee = 37%, conditions: AD-H; hexane/*i*-PrOH = $85/15$; flow rate 0.8 mL/min; $\lambda = 254$ nm; t (major) $= 48.4$ min; t (minor) $= 102.0$ min.

General Procedure for Synthesis of 5. In a round-bottle flask charged with 4a (0.2 mmol) and MeOH (2 mL), NaBH₄ (0.4 mmol, 2.0 equiv) was added. Then, the reaction mixture was stirred vigorously at 25 °C for 1 h until the substrate 4a disappeared (monitored by TLC). At this time, the reaction was diluted with H_2O (10 mL) and exacted with Et₂O (3 \times 10 mL). The combined organic layers were washed with brine (1×10 mL), dried over Na₂SO₄, and filtered. The solvent was removed under reduced pressure, and the crude product was purified by flash chromatography (the crude residue was dry loaded with silica gel, 1/20 to 1/4 ethyl acetate/petroleum ether) to provide product 5 as a white solid (0.058 g, 82%). Mp: 73− 74 °C; ¹ H NMR (400 MHz, CDCl3): δ 8.52−8.50 (m, 1H), 7.58− 7.56 (m, 1H), 7.45−7.43 (m, 1H),7.39−7.37 (m, 1H), 7.30−7.21 (m, 4H), 7.74−7.69 (m, 3H), 6.47 (d, J = 7.2 Hz, 2H), 4.67 (d, J = 2.4 Hz, 1H), 4.27−4.21 (m, 1H), 3.54 (brs, 1H), 3.28−3.26 (m, 1H), 3.21− 3.17 (m, 1H), 3.10−3.06 (m, 1H), 2.54−2.53 (m, 1H), 2.33−2.30 (m, 1H), 2.19 (dd, J = 13.6 Hz, 2.4 Hz, 1H); 13C NMR (125 MHz, CDCl3): δ 149.4, 149.1, 145.8, 140.7, 140.3, 128.8, 128.3, 128.1, 127.4, 127.1, 126.9, 126.5, 124.7, 123.7, 119.7, 119.2, 72.9, 67.4, 62.6, 49.2, 45.3, 33.6; IR (thin film) 3399, 3058, 2925, 2865, 1593, 1490, 1094, 803, 745 cm⁻¹; HRMS (ESI) m/z calcd for $C_{24}H_{24}NO_2$ $(M + H)^+$ 358.1807, found 358.1797.

General Procedure for Synthesis of 6. To a solution of Dess− Martin periodinane (0.064 g, 0.15 mmol) in CH_2Cl_2 (1.0 mL) was added a solution of 4a $(0.036 \text{ g}, 0.1 \text{ mmol})$ in CH_2Cl_2 (1.0 mL) with stirring overnight at room temperature. Once 4a disappeared completely (about 15 h), the reaction mixture was diluted with diethyl ether (5 mL). Aqueous 1.0 M NaOH (2 mL) was added, and the solution was stirred for a further 20 min. The organic phase was then separated, and the aqueous phase was extracted with Et₂O (3×5) mL). The combined organic phases were dried over $Na₂SO₄$ and concentrated in vacuo. The crude product was purified by flash chromatography (the crude residue was dry loaded with silica gel, 1/20 to 1/4 ethyl acetate/petroleum ether) to provide product 6 as a light yellow solid (0.031 g, 88%). Mp: 176−177 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.63 (d, J = 7.2 Hz, 2H), 7.54 (d, J = 7.2 Hz, 2H), 7.28– 7.16 (m, 6H), 7.01−6.96 (m, 2H), 6.76 (d, J = 8.0 Hz, 2H), 6.37 (brs, 1H), 3.87 (d, J = 7.6 Hz, 2H), 2.72−2.68 (m, 2H); 13C NMR (100 MHz, CDCl₃): δ 193.7, 149.0, 147.1, 144.8, 139.3, 129.5, 128.8, 128.2, 124.6, 124.5, 120.0, 118.2, 114.3, 83.3, 47.2, 36.3; IR (thin film) 3328, 3044, 2924, 2857, 1734, 1574, 1295, 743, 687 cm⁻¹; HRMS (ESI) m/z calcd for $C_{24}H_{18}NO (M–H_2O)^+$ 336.1388, found 336.1371.

General Procedure for Synthesis of Nitrone 1. A round-bottle flask, open to the air, was charged with fluoren-9-one oxime (0.5 mmol), CCl_4 (5 mL), and KOH (0.75 mmol, 1.5 equiv). The mixture was stirred vigorously at room temperature for 5 min. Then, diaryliodonium salt (0.75 mmol, 1.5 equiv) was added in one portion. The reaction was monitored by TLC until the oxime was consumed completely. At this time, the $CCI₄$ was removed under reduced pressure, and the crude product was purified by flash chromatography (the crude residue was dry loaded on silica gel; 1/10−1/4 ethyl acetate/petroleum ether) to provide nitrones 1 as yellow solid.

The nitrones $1a_1^{9d} 1b_1^{13} 1d_1^{13} 1g_1^{13} 1f_1^{14} 1n_1^{15} 1o_1^{16}$ and $1p^{17}$ were prepared according to literature methods, and their spectral data matched literature [va](#page-9-0)lues.

N-(9H-Fluoren-9-ylid[ene](#page-9-0))-4[-p](#page-9-0)he[nox](#page-9-0)y[ani](#page-9-0)line [o](#page-9-0)xid[e](#page-9-0) (1c). A [y](#page-9-0)ellow solid, 0.040 g, 37% yield. Mp: 178–179 °C; ¹H NMR (400 MHz; CDCl₃): δ 8.84 (d, J = 7.2 Hz, 1H), 7.62–7.56 (m, 2H), 7.42–7.39 (m, 3H), 7.36−7.32 (m, 3H), 7.20−7.17 (m, 2H), 7.14−7.04 (m, 5H),

6.91 (t, J = 7.6 Hz, 1H), 6.05 (d, J = 8.0 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 159.0, 155.9, 145.6, 142.0, 139.2, 139.0, 132.3, 131.2, 130.8, 130.0, 129.2, 128.8, 127.2, 127.0, 125.5, 124.4, 123.8, 120.2, 119.7, 119.6, 119.2; IR (thin film) 3058, 1586, 1487, 1245, 870, 727 cm⁻¹; HRMS (ESI) m/z calcd for C₂₅H₁₈NO₂ (M + H)⁺ 364.1338, found 364.1321.

N-(9H-Fluoren-9-ylidene)-4-isopropylaniline oxide (1e). A yellow solid, 0.040 g, 43% yield. Mp: 95−96 °C; ¹ H NMR (500 MHz; CDCl₃): δ 8.85 (d, J = 7.6 Hz, 1H), 7.61 (d, J = 7.6 Hz, 1H), 7.55 (d, J $= 7.6$ Hz, 1H), 7.40–7.31 (m, 6H), 7.16 (t, J = 7.6 Hz, 1H), 6.83–6.79 $(m, 1H)$, 5.87 (d, J = 8.4 Hz, 1H), 2.98–2.94 $(m, 1H)$, 1.25 (d, J = 6.8) Hz, 6H); ¹³C NMR (125 MHz, CDCl₃): δ 151.4, 145.3, 144.9, 139.1, 139.0, 132.3, 131.0, 130.8, 129.0, 128.7, 127.9, 127.1, 126.9, 123.8, 123.5, 120.0, 119.5, 33.9, 23.8; IR (thin film) 3054, 2957, 1534, 1498, 1253, 828, 730 cm⁻¹; HRMS (ESI) m/z calcd for C₂₂H₂₀NO (M + H)+ 314.1545, found 314.1534.

N-(9H-Fluoren-9-ylidene)-4-(methoxycarbonyl)aniline oxide (1h). A yellow solid, 0.042 g, 25% yield. Mp: 109−110 °C; ¹H NMR (400 MHz, CDCl₃): δ 8.83 (d, J = 7.6 Hz, 1H), 8.22 (d, J = 8.4 Hz, 2H), 7.63 (d, J = 7.2 Hz, 1H), 7.58−7.53 (m, 3H), 7.45−7.41 (m, 1H), 7.37 $(t, J = 7.6 \text{ Hz}, 1H), 7.20-7.16 \text{ (m, 1H)}, 6.84 \text{ (t, } J = 7.6 \text{ Hz}, 1H), 5.85$ (d, J = 8.0 Hz 1H), 3.93 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 165.7, 149.7, 145.9, 139.4, 139.1, 134.6, 132.1, 131.8, 131.6, 131.5, 130.3, 129.5, 128.9, 127.4, 127.1, 124.0, 123.7, 120.3, 119.7, 52.6; IR (thin film) 3055, 2953, 1718, 1537, 1441, 1282, 952, 727 cm⁻¹; HRMS (ESI) m/z calcd for $C_{21}H_{16}NO_3 (M + H)^+$ 330.1130, found 330.1116.

N-(9H-Fluoren-9-ylidene)-3,5-dimethylaniline oxide (1i). A yellow solid, 0.050 g, 56% yield. Mp: 171–172 °C; ¹H NMR (400 MHz; CDCl₃): δ 8.85 (d, J = 7.6 Hz, 1H), 7.63 (d, J = 7.2 Hz, 1H), 7.58 (d, J $= 7.6$ Hz, 1H), 7.43 (t, J = 7.2 Hz, 1H), 7.37 (t, J = 7.6 Hz, 1H), 7.20− 7.15 (m, 2H), 7.05 (s, 2H), 6.88 (t, $J = 7.6$ Hz, 1H), 5.93 (d, $J = 7.6$ Hz, 1H), 2.33 (s, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 146.9, 145.2. 140.2, 139.1, 139.0, 132.3, 131.7, 131.0, 130.8, 129.0, 128.8, 127.2, 127.0, 124.0, 121.1, 120.0, 119.5, 21.2; IR (thin film) 3036, 2917, 1533, 1445, 1246, 768, 722 cm[−]¹ ; HRMS (ESI) m/z calcd for $C_{21}H_{18}NO (M + H)^+$ 300.1388, found 300.1375.

N-(9H-Fluoren-9-ylidene)-3-nitroaniline oxide (1j). A yellow solid, 0.067 g, 42% yield. Mp: 150−151 °C; ¹H NMR (400 MHz, CDCl₃): δ 8.89 (d, J = 7.6 Hz, 1H), 8.51−8.45 (m, 2H), 7.92 (d, J = 8.0, 1H), 7.85−7.81 (m, 1H), 7.72−7.66 (m, 2H), 7.55−7.51 (m, 1H), 7.46 (t, J $= 7.6$ Hz, 1H), $7.31 - 7.26$ (m, 1H), 6.93 (t, J = 7.6 Hz, 1H), 5.93 (d, J $= 8.0$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 148.9, 147.2, 146.3, 139.6, 139.2, 131.9, 131.8, 131.3, 130.0, 129.8, 129.1, 127.4, 127.2, 125.0, 123.2, 120.6, 119.9, 119.8; IR (thin film) 3055, 1600, 1531, 1447, 1346, 1250, 727 cm⁻¹; HRMS (ESI) m/z calcd for $C_{19}H_{13}N_2O_3$ $(M + H)^+$ 317.0926, found 317.0912.

2-Bromo-N-(9H-fluoren-9-ylidene)aniline oxide (1k). A yellow solid, 0.104 g, 59% yield. Mp: 62−63 °C; ¹ H NMR (400 MHz, CDCl₃): δ 8.87 (d, J = 7.2 Hz, 1H), 7.74 (d, J = 8.0 Hz, 1H), 7.63– 7.56 (m, 2H), 7.49−7.33 (m, 5H), 7.21−7.17 (m, 1H), 6.86−6.82 (m, 1H), 5.71 (d, J = 8.0 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 146.3, 145.8, 139.4, 139.3, 134.3, 131.7, 131.5, 131.3, 130.2, 129.5 129.3, 128.9, 127.6, 127.4, 125.6, 122.9, 120.3, 119.6, 117.0; IR (thin film) 3061, 1601, 1541, 1451, 1255, 768, 729 cm[−]¹ ; HRMS (ESI) m/z calcd for $C_{19}H_{13}BrNO (M + H)^+$ 350.0181, found 350.0167.

N-(2,7-Dibromo-9H-fluoren-9-ylidene)aniline oxide (1l). A yellow solid, 0.146 g, 57% yield. Mp: 220−221 °C; ¹ H NMR (400 MHz, CDCl₃): δ 9.06 (s, 1H), 7.66–7.60 (m, 4H), 7.52–7.45 (m, 4H), 7.37 (d, $J = 8.0$ Hz, 1H), 5.92 (s, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 146.6, 143.4, 136.7, 136.6, 133.9, 133.5, 132.1, 131.9, 130.8, 130.3, 129.6, 127.0, 123.6, 123.5, 123.0, 121.3, 120.8; IR (thin film) 3022, 1536, 1432, 1251, 815, 765, 686 cm[−]¹ ; HRMS (ESI) m/z calcd for $C_{19}H_{12}Br_2NO (M + H)^+$ 427.9286, found 427.9280.

 $N-(2\text{-}\mathsf{Bromo}\text{-}\mathsf{9H}\text{-}\mathsf{fluoren}\text{-}\mathsf{9-ylidene})$ aniline oxide (1m). $E/Z = 1/1$, a yellow solid, 0.090 g, 43% yield, mp: 136−137 °C. Isomer 1: ¹H NMR (400 MHz, CDCl₃): δ 9.09 (s, 1H), 7.66–7.62 (m, 5H), 7.53– 7.50 (m, 4H), 7.35 (d, J = 8.0 Hz, 1H), 5.91−5.88 (m, 1H); 13C NMR $(100 \text{ MHz}, \text{CDCl}_3): \delta$ 146.9, 144.5, 138.1, 137.7, 133.8, 132.3, 131.7, 130.6, 130.5, 130.2, 129.6, 129.2, 126.9, 123.7, 122.5, 121.2, 120.2. Isomer 2: ¹H NMR (400 MHz, CDCl₃): δ 8.89 (d, J = 7.6 Hz, 1H),

7.61−7.58 (m, 4H), 7.49−7.41 (m, 4H), 7.25−7.21 (m, 1H), 6.93 (t, J $= 7.6$ Hz, 1H), 5.91–5.88 (m, 1H); ¹³C NMR (100 MHz, CDCl₃): δ 146.7, 144.4, 138.0, 137.6, 133.7, 132.0, 131.2, 130.4, 130.1, 129.1, 127.5, 126.8, 123.6, 123.5, 120.9, 120.7, 119.6; IR (thin film) 3054, 1603, 1534, 1440, 1345, 1252, 771 cm[−]¹ ; HRMS (ESI) m/z calcd for $C_{19}H_{13}BrNO (M + H)^+$ 350.0181, found 350.0172.

General Procedure for Synthesis of Methylenecyclopropane 2. Bromoacetyl bromide (2.61 mL, 30.0 mmol) was added dropwise to a solution of corresponding alcohol (30 mmol) and pyridine (2.4 mL, 30 mmol) in CH₂Cl₂ (40 mL) at 0 $^{\circ}$ C to form a white suspension that was stirred for 20 min at 0 °C and then an additional 30 min at 25 °C. At this time, H_2O (50 mL) was added to the reaction mixture, and the organic layer was separated. The aqueous layer was then extracted with $CH₂Cl₂$ (2 × 20 mL). The combined organic layers were washed with brine (50 mL) and dried over MgSO₄. The filtrate was then concentrated to give bromoester as a colorless liquid which was used directly without purification for the next step.

Bromoester was added dropwise to a toluene (100 mL) solution of triphenylphosphine (7.8 g, 30 mmol) to form a white precipitate. The resulting slurry was stirred overnight, filtered over a glass-fitted filter, and washed with toluene and hexane to give phosphonium salt as a white solid. The phosphonium salt was dissolved in H_2O (100 mL), and NaOH $(2 M)$ was added to keep the aqueous pH > 7.0. A white solid appeared, and the mixture was stirred for 30 min before CH_2Cl_2 (50 mL) was added. The organic layer was separated and washed with brine (50 mL) and dried over $MgSO₄$. The filtrate was concentrated to give ylide as a white solid, which was used directly without further purification for the next step.

To a refluxing solution of ylide (10 mmol) and benzoic acid (122 mg, 10 mmol %) in benzene (50 mL) was added (1 ethoxycyclopropoxy)trimethylsilane (2 mL, 10 mmol) via syringe. After the addition was completed, the mixture was allowed to reflux at 80 °C for 18 h. After the mixture was cooled to room temperature, benzene was removed under reduced pressure, and the crude product was purified by flash chromatography (the crude residue was dry loaded on silica gel; 1/100−1/20 ethyl acetate/petroleum ether) to afford methylenecyclopropane 2 as a colorless oil.

The methylenecyclopropane $2d¹⁸$ $2e¹⁹$ $2f¹⁹$ $2g¹⁹$ $2h²⁰$ $2i²¹$ $2j²²$ and $2k^{23}$ were prepared according to literature methods, and their spectral data matched literature va[lue](#page-9-0)s.

Cinn[am](#page-9-0)yl-2-cyclopropylideneacetat[e](#page-9-0) [\(](#page-9-0)2a). [A](#page-9-0) co[lor](#page-9-0)les[s](#page-9-0) [o](#page-9-0)il, [1.4](#page-9-0)90 [g,](#page-9-0) 69% yield. ¹H NMR (400 MHz; CDCl₃): δ 7.33 (d, J = 7.2 Hz, 2H), 7.27−7.23 (m, 2H), 7.20−7.16 (m, 1H), 6.63 (d, J = 16.0, 1H), 6.30− 6.21 (m, 2H), 4.77 (d, J = 6.0, 2H), 1.44–1.39 (m, 2H), 1.21–1.16 (m, 2H); 13C NMR (100 MHz, CDCl3): δ 165.8, 145.7, 136.3, 133.7, 128.5, 127.9, 126.5, 123.5, 110.7, 64.7, 4.6, 2.1; IR (thin film) 3031, 2988, 2881, 1713, 1601, 1494, 1170, 741, 695 cm[−]¹ ; HRMS (EI) m/z calcd for $C_{14}H_{14}O_2Na$ $(M + Na)^+$ 237.0891, found 237.0887.

(E)-2-Methyl-3-phenylallyl-2-cyclopropylideneacetate (2b). A colorless oil, 0.800 g, 35% yield. ¹H NMR (400 MHz; CDCl₃): δ 7.36– 7.32 (m, 2H), 7.29−7.23 (m, 3H), 6.57 (s, 1H), 6.30 (s, 1H), 4.73 (s, 2H), 1.92 (s, 3H), 1.48−1.46 (m, 2H), 1.28−1.24 (m, 2H); 13C NMR (100 MHz, CDCl3): δ 165.8, 145.5, 137.1, 133.0, 128.9, 128.1, 127.7, 126.6, 110.8, 69.6, 15.5, 4.6, 2.2; IR (thin film) 3056, 2981, 2872, 1716, 1600, 1491, 1254, 746, 699 cm[−]¹ ; HRMS (EI) m/z calcd for $C_{15}H_{16}O_2$ Na $(M + Na)^+$ 251.1048, found 251.1045.

3-Phenylprop-2-ynyl-2-cyclopropylideneacetate (2c). A colorless oil, 1.200 g, 57% yield. ¹H NMR (400 MHz; CDCl₃): δ 7.46–7.44 (m, 2H), 7.32−7.26 (m, 3H), 6.30 (s, 1H), 5.00 (s, 2H), 1.52−1.48 (m, 2H), 1.30−1.24 (m, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 165.2, 146.8, 131.8, 128.6, 128.2, 122.2, 110.1, 86.1, 83.2, 52.4, 4.7, 2.2; IR (thin film) 3060, 2985, 2875, 1718, 1600, 1490, 1252, 759, 692 cm[−]¹ ; HRMS (EI) m/z calcd for C₁₄H₁₂O₂Na (M + Na) ⁺ 235.0735, found 235.0732.

■ ASSOCIATED CONTENT

6 Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b02544.

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Notes

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■ REFERENCES

(1) Some studies on the bioactivity of piperidinones: (a) Das, U.; Pati, H. N.; Baráth, Z.; Csonka, Á.; Molnár, J.; Dimmock, J. R. *Bioorg*. Med. Chem. Lett. 2016, 26, 1319. (b) Cheng, D. H.; Valente, S.; Castellano, S.; Sbardella, G.; Di Santo, R.; Costi, R.; Bedford, M. T.; Mai, A. J. Med. Chem. 2011, 54, 4928. (c) Das, U.; Pati, H. N.; Sakagami, H.; Hashimoto, K.; Kawase, M.; Balzarini, J.; De Clercq, E.; Dimmock, J. R. J. Med. Chem. 2011, 54, 3445. (d) Michelsen, K.; Jordan, J. B.; Lewis, J.; Long, A. M.; Yang, E.; Rew, Y.; Zhou, J.; Yakowec, P.; Schnier, P. D.; Huang, X.; Poppe, L. J. Am. Chem. Soc. 2012, 134, 17059.

(2) Some transformations for piperidinones: (a) Peng, Z. H.; Wong, J. W.; Hansen, E. C.; Puchlopek-Dermenci, A. L. A.; Clarke, H. J. Org. Lett. 2014, 16, 860. (b) Lee, C. W.; Lira, R.; Dutra, J.; Ogilvie, K.; O'Neill, B. T.; Brodney, M.; Helal, C.; Young, J.; Lachapelle, E.; Sakya, S.; Murray, J. C. J. Org. Chem. 2013, 78, 2661. (c) Ellis, G. L.; Amewu, R.; Sabbani, S.; Stocks, P. A.; Shone, A.; Stanford, D.; Gibbons, P.; Davies, J.; Vivas, L.; Charnaud, S.; Bongard, E.; Hall, C.; Rimmer, K.; Lozanom, S.; Jesus, M.; Gargallo, D.; Ward, S. A.; O'Neill, P. M. J. Med. Chem. 2008, 51, 2170. (d) Ramanathan, S. K.; Keeler, J.; Lee, H. L.; Reddy, D. S.; Lushington, G.; Aube, J. Org. Lett. 2005, 7, 1059.

(3) Some reviews on methylenecyclopropanes: (a) Brandi, A.; Cicchi, S.; Cordero, F. M.; Goti, A. Chem. Rev. 2003, 103, 1213. (b) Goti, A.; Cordero, F. M.; Brandi, A. Top. Curr. Chem. 1996, 178, 1. (c) Shi, M.; Lu, J. M.; Shao, L. X. Acc. Chem. Res. 2012, 45, 641. Some recent examples of methylenecyclopropanes: (d) Li, S. Y.; Luo, Y.; Wu, J. Org. Lett. 2011, 13, 3190. (e) Zhu, Z. Z.; Chen, K.; Yu, L. Z.; Tang, X. Y.; Shi, M. Org. Lett. 2015, 17, 5994. (f) Yuan, Y. C.; Yang, H. B.; Tang, X. Y.; Wei, Y.; Shi, M. Chem. - Eur. J. 2016, 22, 5146. (g) Sakae, R.; Matsuda, N.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2014, 16, 1228. (h) Kippo, T.; Hamaoka, K.; Ryu, I. J. Am. Chem. Soc. 2013, 135, 632.

(4) For reviews on the 1,3-cycloaddition of nitrones, see: (a) Shi, W. M.; Ma, X. P.; Su, G. F.; Mo, D. L. Org. Chem. Front. 2016, 3, 116. (b) Anderson, L. L. Asian J. Org. Chem. 2016, 5, 9. (c) Jones, R. C. F.; Martin, J. N. In Synthetic Application of 1,3-Dipolar Cycloaddition Chemistry torward Heterocycles and Natural Products; Padwa, A., Pearson, W. H., Eds.; Wiley: New York, 2002; pp 1. Some recent examples of 1,3-cycloaddition of nitrones:. (d) Barber, J. S.; Styduhar, E. D.; Pham, H. V.; Mcmahon, T. C.; Houk, K. N.; Garg, N. K. J. Am. Chem. Soc. 2016, 138, 2512. (e) Chen, M. J.; Zhang, Z. M.; Yu, Z. Z.;

Qiu, H. L.; Wu, H. H.; Zhang, J. L. ACS Catal. 2015, 5, 7488. (f) Hoogenboom, J.; Zuilhof, H.; Wennekes, T. Org. Lett. 2015, 17, 5550. (g) Molchanov, A. P.; Savinkov, R. S.; Stepakov, A. V.; Starova, G. L.; Kostikov, R. R.; Barnakova, V. S.; Ivanov, A. V. Synthesis 2014, 46, 771.

(5) (a) Brandi, A.; Guarna, A.; Goti, A.; De Sarlo, F. Tetrahedron Lett. 1986, 27, 1727. (b) Hassner, A.; Stumer, C. Organic Syntheses based on Name Reactions: Elsevier: Pergamon, 2002; pp 42. (c) Brandi, A.; Garro, S.; Guarna, A.; Goti, A.; Cordero, F. M.; De Sarlo, F. J. Org. Chem. 1988, 53, 2430. (d) Brandi, A.; Durust, Y.; Cordero, F. M.; De Sarlo, F. J. Org. Chem. 1992, 57, 5666. (e) Cordero, F. M.; Brandi, A.; Querci, C.; Goti, A.; De Sarlo, F.; Guarna, A. J. Org. Chem. 1990, 55, 1762. (f) Brandi, A.; Cordero, F. M.; Goti, A.; De Sarlo, F.; Guarna, A. Synlett 1993, 1993, 1. (g) Cordero, F. M.; De Sarlo, F.; Brandi, A. Monatsh. Chem. 2004, 135, 649.

(6) Cordero, F. M.; Goti, A.; De Sarlo, F.; Guarna, A.; Brandi, A. Tetrahedron 1989, 45, 5917.

(7) Tran, T. Q.; Diev, V. V.; Starova, G. L.; Gurzhiy, V. V.; Molchanov, A. P. Eur. J. Org. Chem. 2012, 2012, 2054.

(8) (a) Shi, W. M.; Ma, X. P.; Pan, C. X.; Su, G. F.; Mo, D. L. J. Org. Chem. 2015, 80, 11175. (b) Ma, X. P.; Shi, W. M.; Mo, X. L.; Li, X. H.; Li, L. G.; Pan, C. X.; Chen, B.; Su, G. F.; Mo, D. L. J. Org. Chem. 2015, 80, 10098. (c) Wang, Z. X.; Shi, W. M.; Bi, H. Y.; Li, X. H.; Su, G. F.; Mo, D. L. J. Org. Chem. 2016, 81, 8014.

(9) Some examples for 9-fluorene nitrones: (a) Natarajan, R.; Rappai, J. P.; Unnikrishnan, P. A.; Radhamani, S.; Prathapan, S. Synlett 2015, 26, 2467. (b) Pecak, W. H.; Son, J.; Burnstine, A. J.; Anderson, L. L. Org. Lett. 2014, 16, 3440. (c) Mo, D. L.; Wink, D. A.; Anderson, L. L. Org. Lett. 2012, 14, 5180. (d) Mugnier, Y.; Gard, J. C.; Huang, Y. Q.; Couture, Y.; Lasia, A.; Lessard, J. J. Org. Chem. 1993, 58, 5329.

(10) A recent review on rearrangement of N−O bond cleavage: Tabolin, A. A.; Ioffe, S. L. Chem. Rev. 2014, 114, 5426.

(11) For examples of the use of fluorene derivatives in pharmaceuticals and materials, see: (a) Hizliates, C. G.; Guelle, S.; Erguen, Y. J. Heterocycl. Chem. 2016, 53, 249. (b) Makanga, M.; Krudsood, S. Malar. J. 2009, 8, S5. (c) Andrade, C. D.; Yanez, C. O.; Rodriguez, L.; Belfield, K. D. J. Org. Chem. 2010, 75, 3975. (d) Kurdyukova, I. V.; Ishchenko, A. A. Russ. Chem. Rev. 2012, 81, 258. (e) Leclerc, M. J. Polym. Sci., Part A: Polym. Chem. 2001, 39, 2867. (f) Cossy, J.; Poitevin, C.; Pardo, D. G. J. Org. Chem. 1998, 63, 4554.

(12) The possible mechanism for formation of product 6 might include oxidation of the hydroxyl group by DMP, followed with a retro aza-Michael addition, intramolecular condensation, and isomerization. See: (a) Mo, D. L.; Wink, D. A.; Anderson, L. L. Chem. - Eur. J. 2014, 20, 13217. (b) Bravo, F.; Cimarosti, Z.; Tinazzi, F.; Smith, G. E.; Castoldi, D.; Provera, S.; Westerduin, P. Org. Process Res. Dev. 2010, 14, 1162.

(13) Abou-Gharbia, M. A.; Joullie, M. M. J. Org. Chem. 1979, 44, 2961.

(14) Huisgen, R.; Fleischmann, R.; Eckell, A. Chem. Ber. 1977, 110, 500.

(15) Abou-Gharbia, M. A.; Joullié, M. M. J. Org. Chem. 1979, 44, 2961.

(16) Nguyen, D. V.; Prakash, P.; Gravel, E.; Doris, E. RSC Adv. 2016, 6, 89238.

(17) Friscourt, F.; Fahrni, C. J.; Boons, G. J. Chem. - Eur. J. 2015, 21, 13996.

(18) Wessjohann, L.; Krass, N.; Yu, D.; De Meijere, A. Chem. Ber. 1992, 125, 867.

(19) Ohashi, M.; Taniguchi, T.; Ogoshi, S. Organometallics 2010, 29, 2386.

(20) Spitzner, D.; Swoboda, H. Tetrahedron Lett. 1986, 27, 1281.

(21) Lechevallier, A.; Huet, F.; Conia, J. M. Tetrahedron 1983, 39, 3317.

(22) Szcześniak, P.; Pieczykolan, M.; Stecko, S. J. Org. Chem. 2016, 81, 1057.

(23) Wang, Y. H.; Muratore, M. E.; Rong, Z. T.; Echavarren, A. M. Angew. Chem., Int. Ed. 2014, 53, 14022.